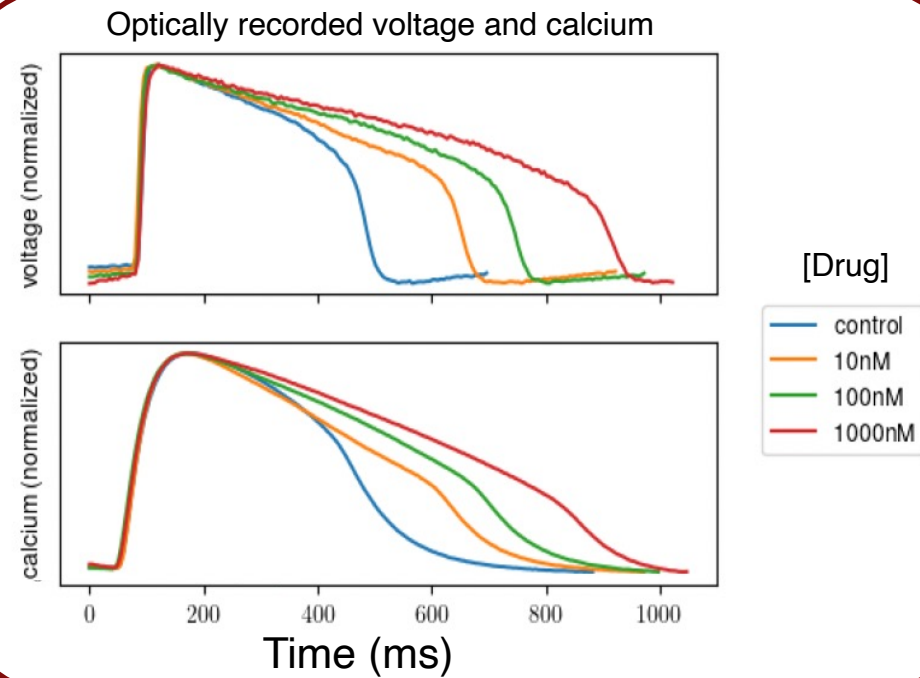
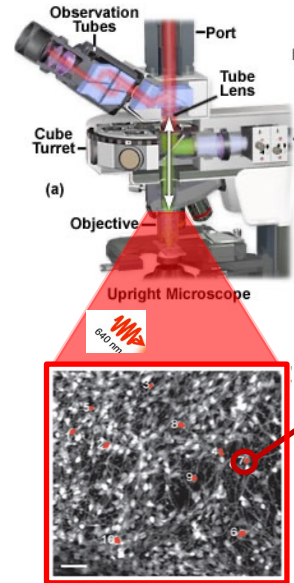


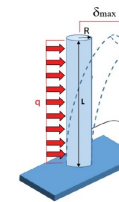
# Cardiac MPS assay measurements

## FLUORESCENCE MICROSCOPY



## MECHANICS

Engineered post deformation

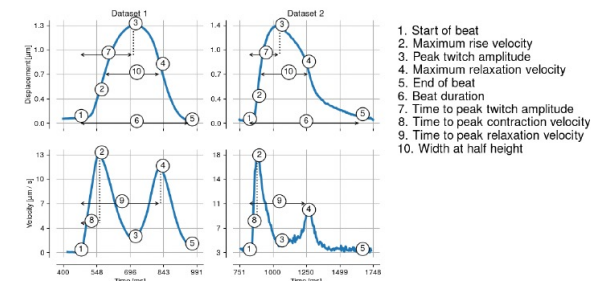
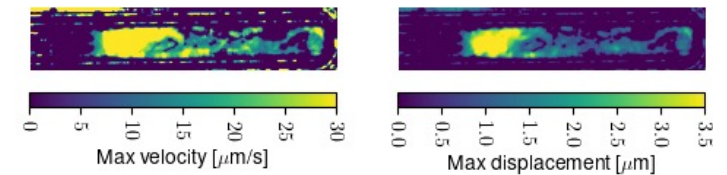


$$F = qL = \frac{8EI\delta_{max}}{L^3}$$

with  $\delta_{max} = \frac{qL^4}{8EI}$

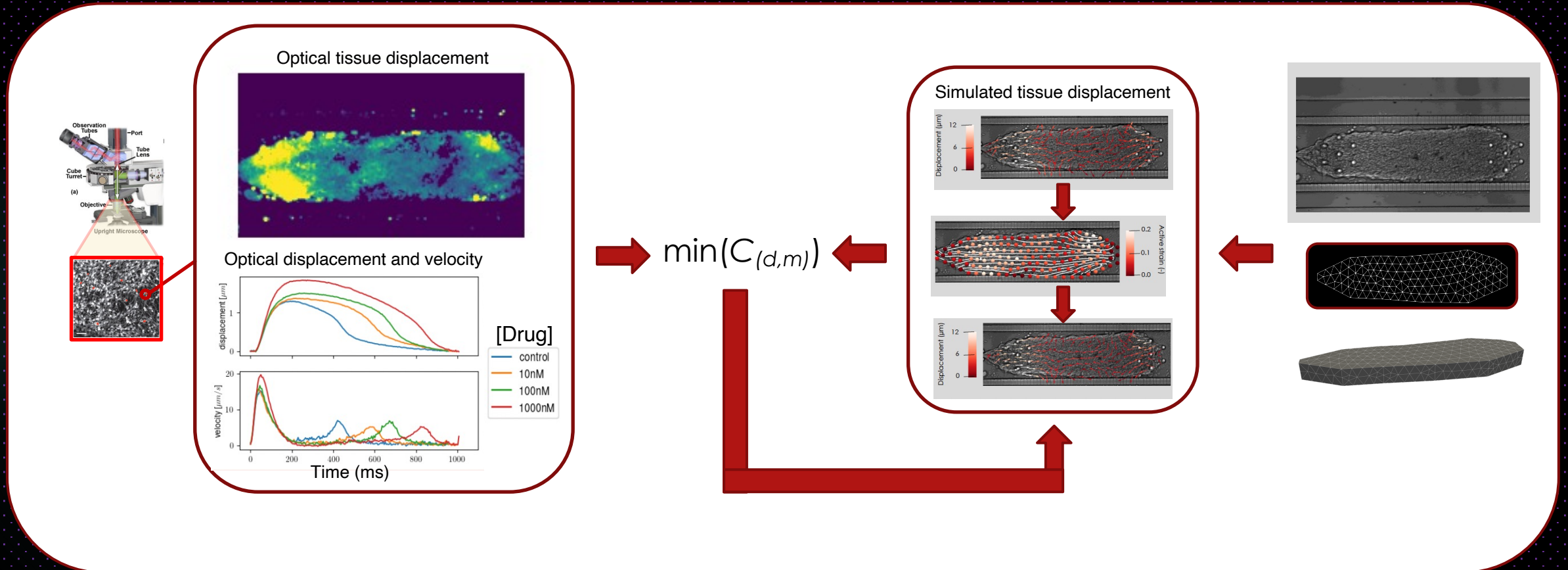
And  $I = \frac{1}{6}\pi R^4$

Pixel tracking



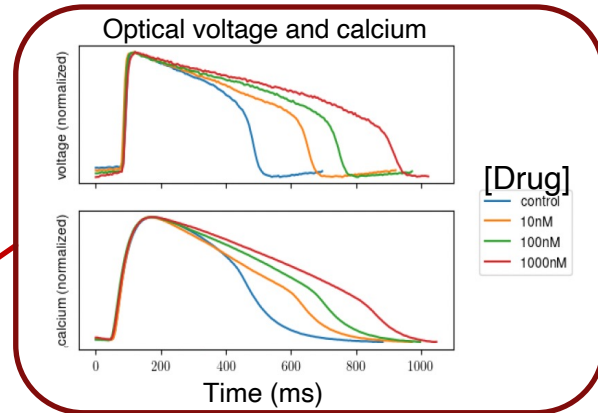
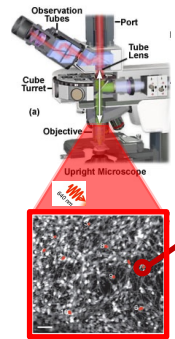
# Measurement constraints in MPS data streams

## ELECTRO-MECHANICAL INVERSE PROBLEM

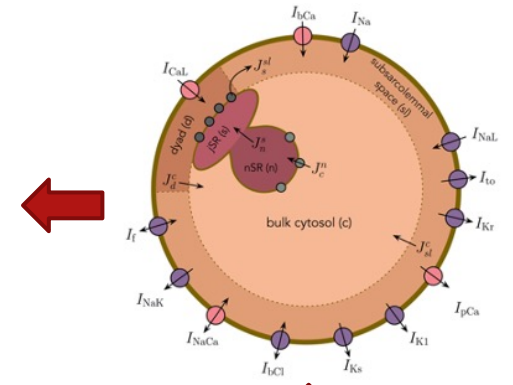
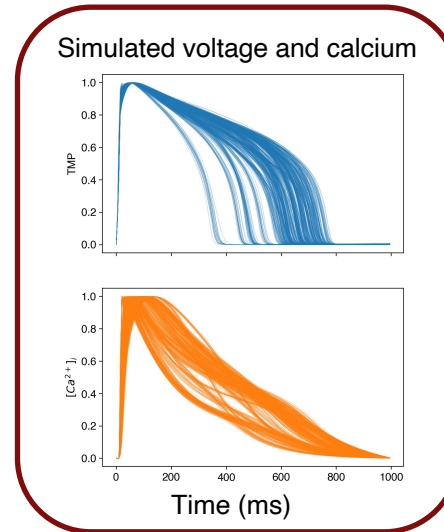


# Measurement constraints in MPS data streams

## ELECTRO-IONIC INVERSE PROBLEM



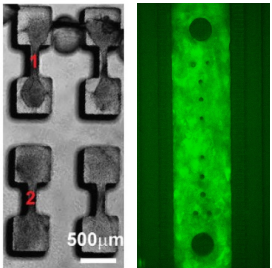
$$\min(C_{(d,m)})$$



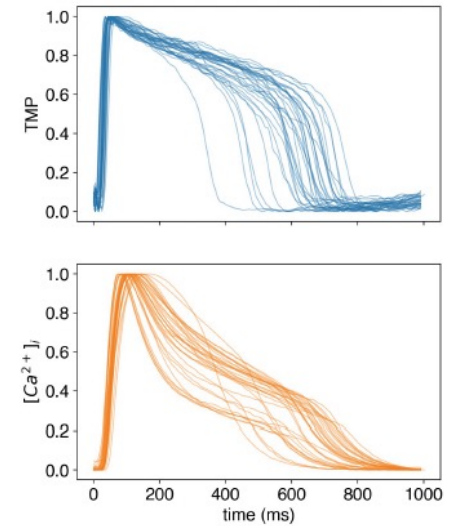
# Measurement constraints in MPS data streams

## SOURCES OF VARIABILITY IN THE MICROTISSUE ASSAYS

Observational  
unit



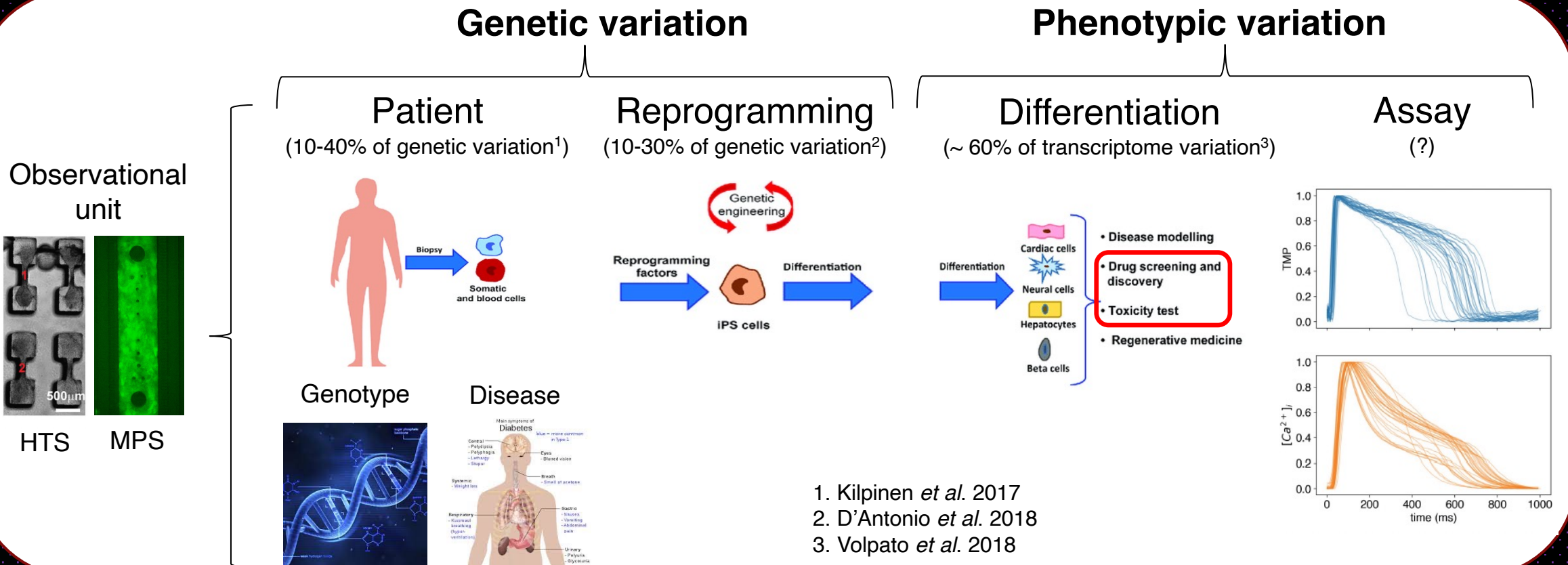
HTS    MPS





# Measurement constraints in MPS data streams

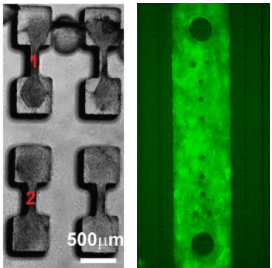
## SOURCES OF VARIABILITY IN THE MICROTISSUE ASSAYS



# Measurement constraints in MPS data streams

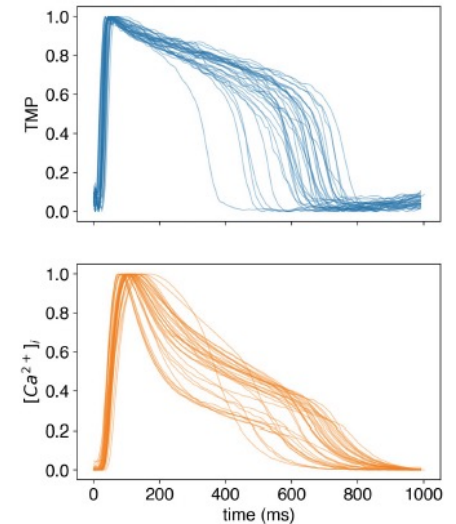
## SOURCES OF VARIABILITY IN THE MICROTISSUE ASSAYS

Observational  
unit



HTS    MPS

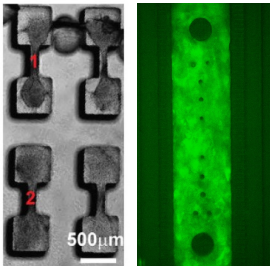
Question: How are we addressing most of this variability through experimental design?



# Measurement constraints in MPS data streams

## SOURCES OF VARIABILITY IN THE MICROTISSUE ASSAYS

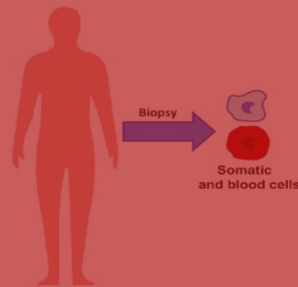
Observational  
unit



HTS MPS

### Genetic variation

**Patient**  
(10-40% of genetic variation<sup>1</sup>)



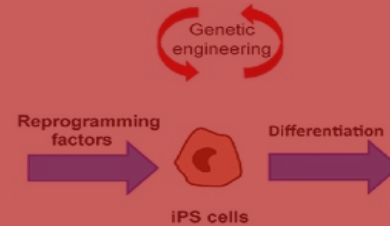
Genotype



Disease

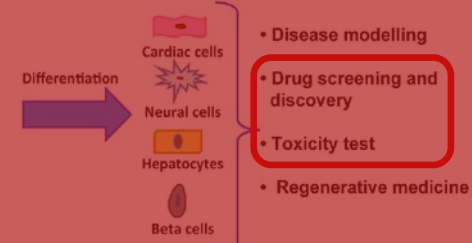


**Reprogramming**  
(10-30% of genetic variation<sup>2</sup>)

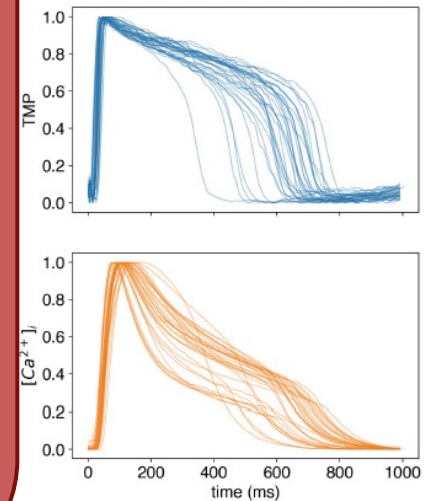


### Phenotypic variation

**Differentiation**  
(~ 60% of transcriptome variation<sup>3</sup>)



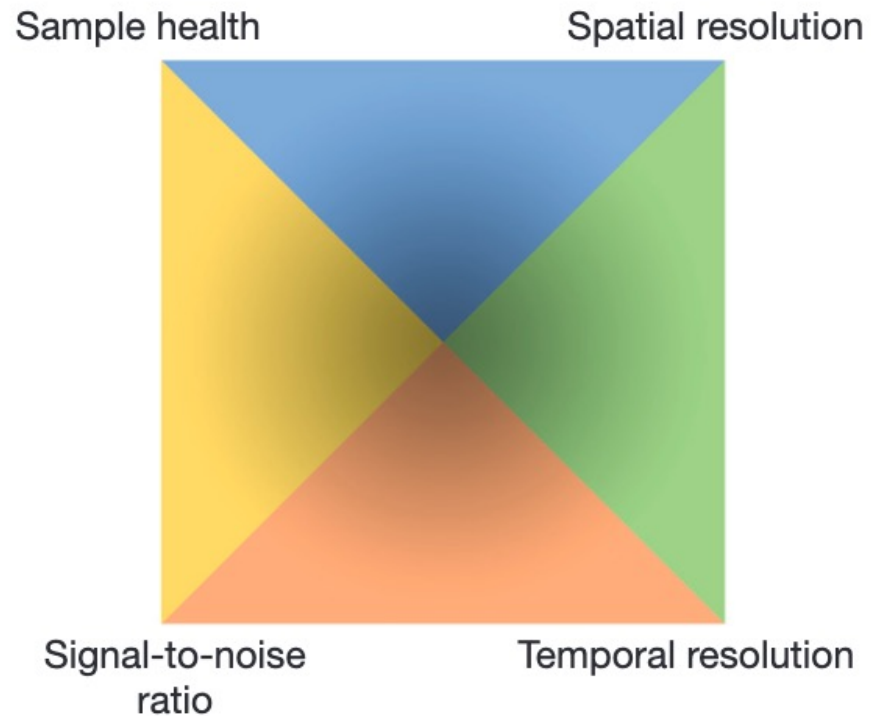
**Assay**  
(?)



1. Kilpinen *et al.* 2017
2. D'Antonio *et al.* 2018
3. Volpato *et al.* 2018

# Measurement constraints in MPS data streams

## THE PYRAMID OF FRUSTRATION IN LIVE CELL MICROSCOPY

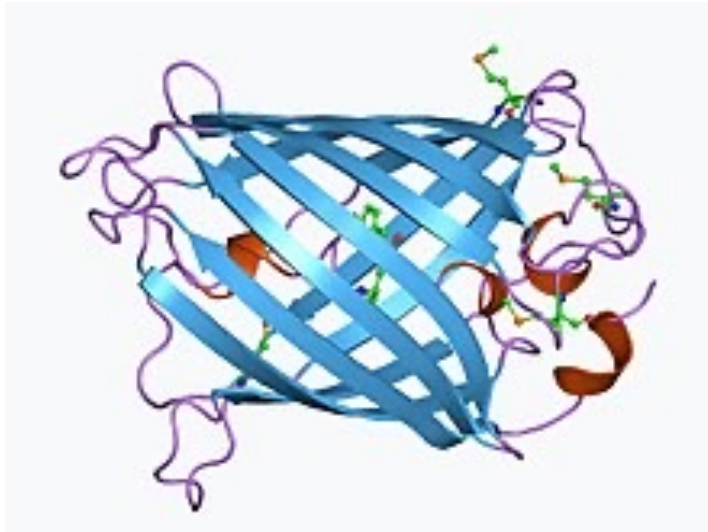


**Figure 2** | The four main considerations for live imaging. This is also known as the 'pyramid of frustration', as no single parameter can be optimized without compromising the others.

# Basis of Fluorescence

## WHAT IS FLUORESCENCE?

### Green Fluorescent Protein - GFP

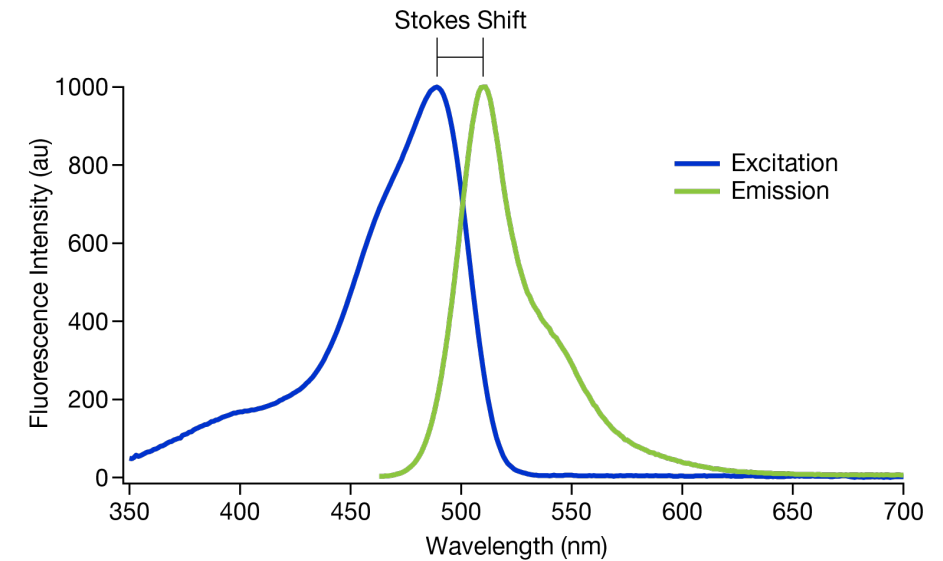
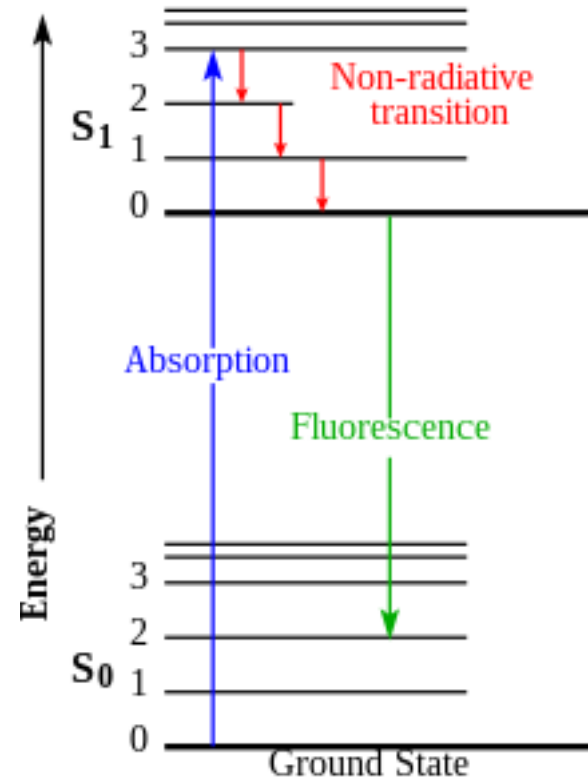
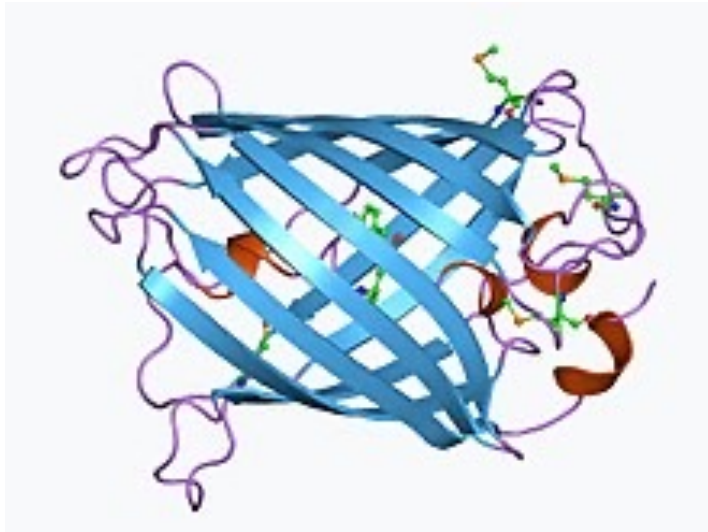




# Basis of Fluorescence

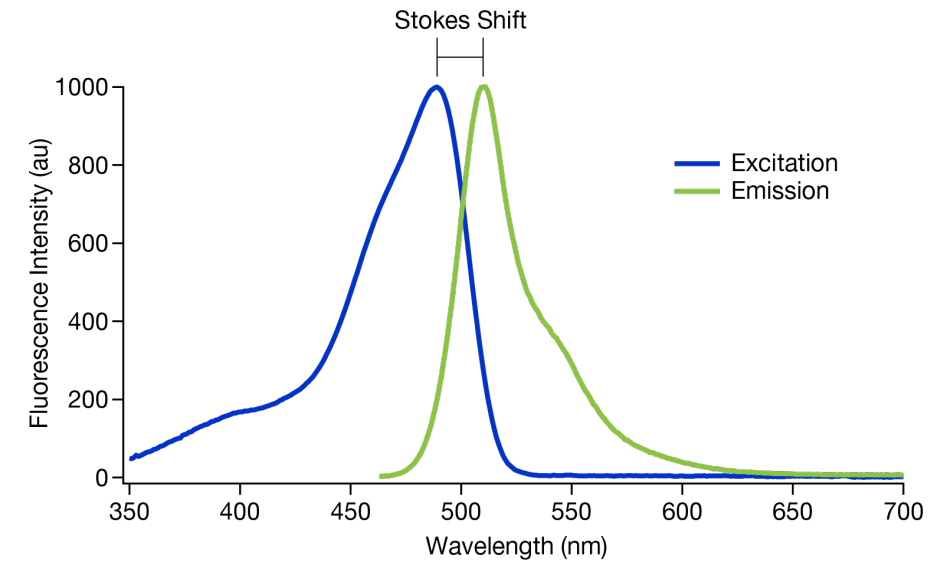
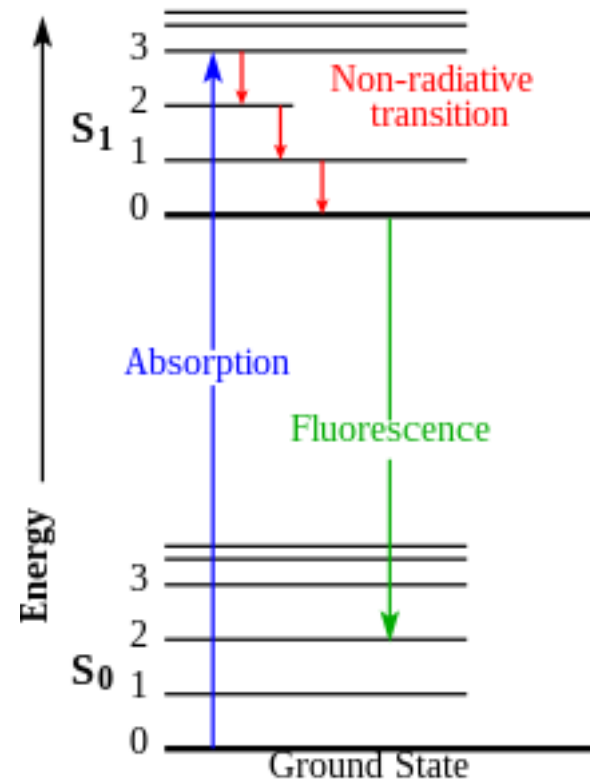
## ENERGY ABSORPTION AND EMISSION

### Green Fluorescent Protein - GFP



# Basis of Fluorescence

## WHAT IS FLUORESCENT QUANTUM YIELD?



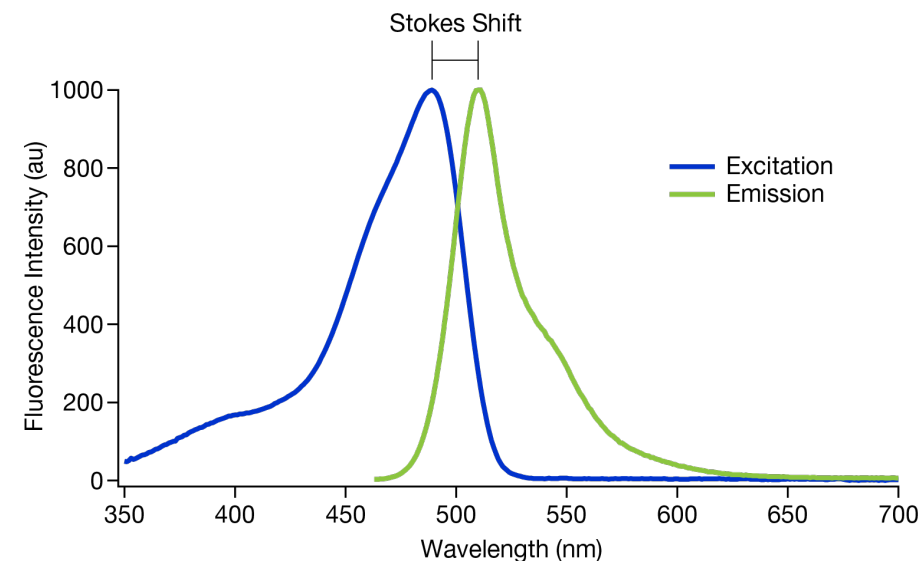
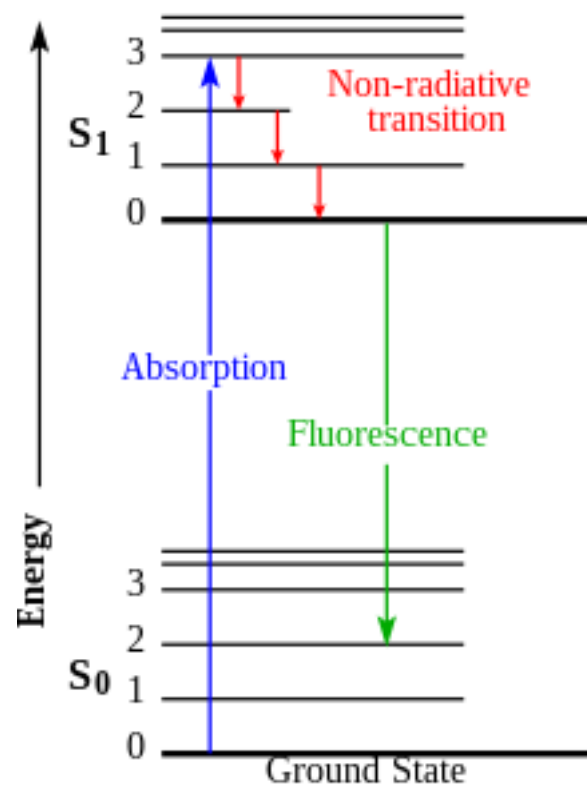
# Basis of Fluorescence

## WHAT IS FLUORESCENT QUANTUM YIELD?

### Quantum yield

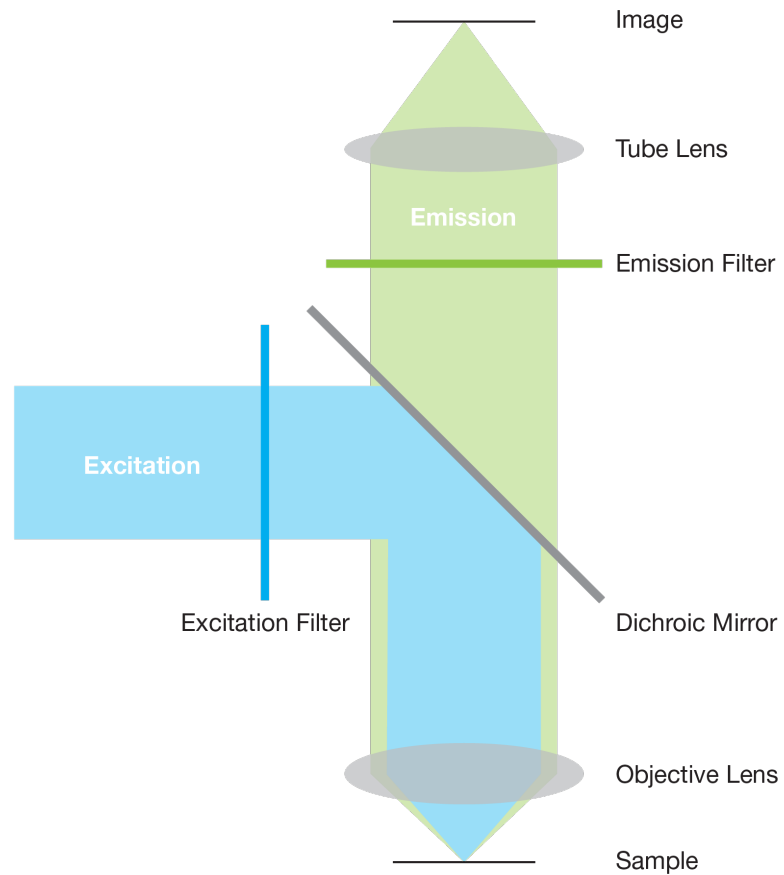
$$\Phi(\lambda) = \frac{\text{number of events}}{\text{number of photons absorbed}}$$

Compound	Solvent	Literature Quantum yield	Emission range / nm
Cresyl violet	Methanol	0.54	600-650
Rhodamine 101	Ethanol + 0.01% HCl	1.00	600-650
Quinine sulfate	0.1M H <sub>2</sub> SO <sub>4</sub>	0.54	400-600
Fluorescein	0.1M NaOH	0.79	500-600
Norharmame	0.1M H <sub>2</sub> SO <sub>4</sub>	0.58	400-550
Harmame	0.1M H <sub>2</sub> SO <sub>4</sub>	0.83	400-550
Harmine	0.1M H <sub>2</sub> SO <sub>4</sub>	0.45	400-550
2-methylharmame	0.1M H <sub>2</sub> SO <sub>4</sub>	0.45	400-550
Chlorophyll A	Ether	0.32	600-750
Zinc phthalocyanine	1% pyridine in toluene	0.30	660-750
Benzene	Cyclohexane	0.05	270-300
Tryptophan	Water, pH 7.2, 25C	0.14	300-380
2-Aminopyridine	0.1M H <sub>2</sub> SO <sub>4</sub>	0.60	315-480
Anthracene	Ethanol	0.27	360-480
9,10-diphenyl anthracene	Cyclohexane	0.90	400-500



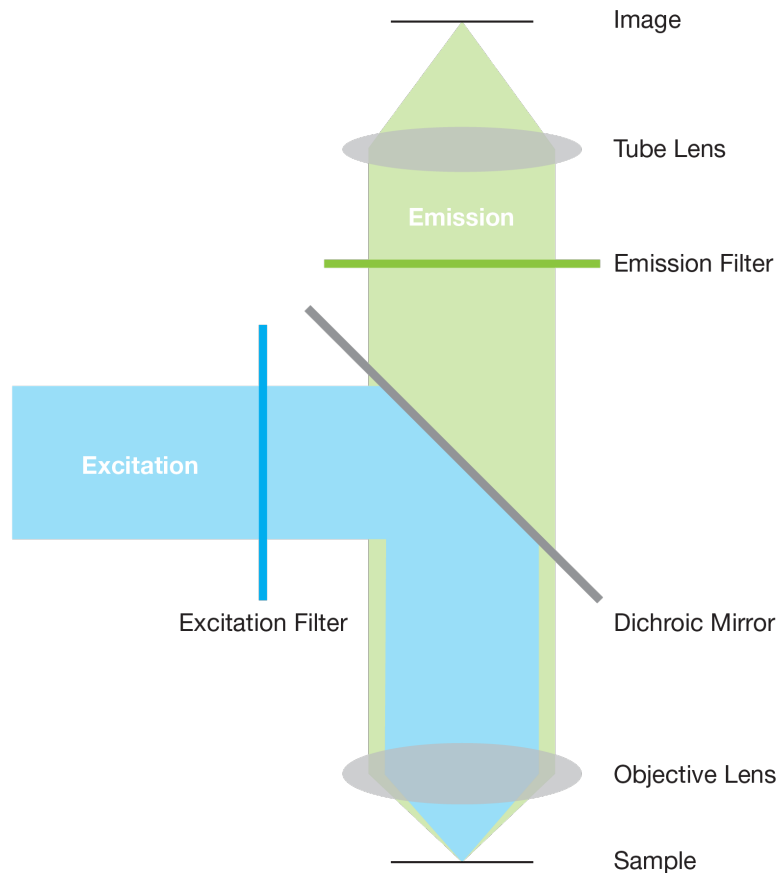
# Fluorescence microscopy

WHAT ARE THE DETERMINANTS OF FLUORESCENCE INTENSITY?



# Fluorescence microscopy

## WHAT ARE THE DETERMINANTS OF FLUORESCENCE INTENSITY?



$$I_f = kI_o\Phi[\epsilon bc]$$

$k$  optical path loss coefficient

$I_o$  incident light intensity

$\Phi$  quantum yield

$\epsilon$  molar absorptivity

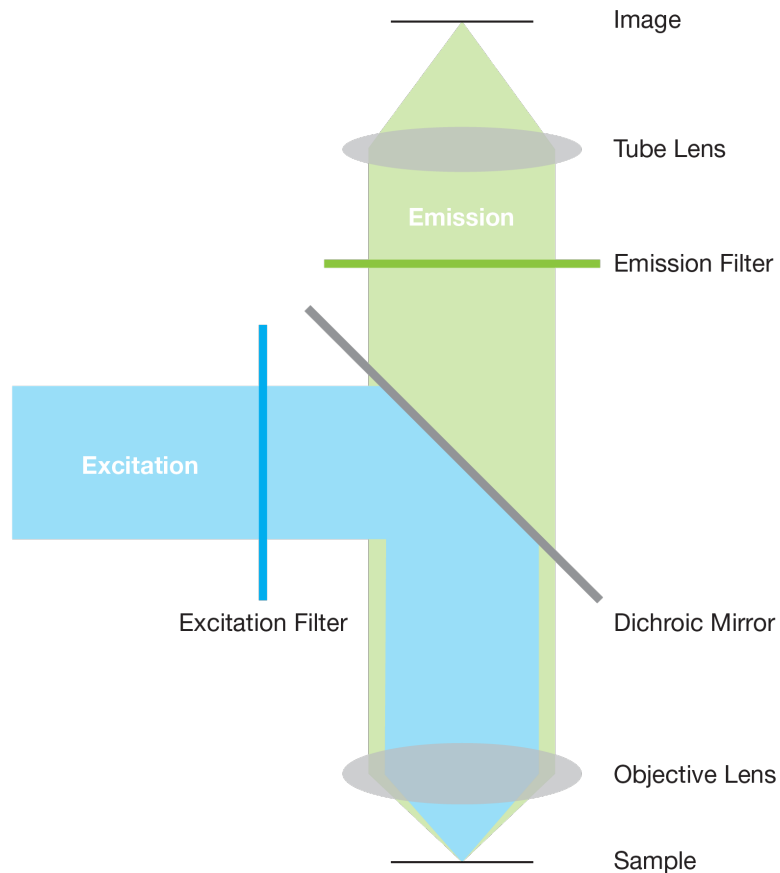
$b$  light path length

$c$  concentration of the fluorophore



# Fluorescence microscopy

## WHAT ARE THE DETERMINANTS OF FLUORESCENCE INTENSITY?



$$I_f = kI_o\Phi[\epsilon bc]$$

$k$  optical path loss coefficient

$I_o$  incident light intensity

$\Phi$  quantum yield

$\epsilon$  molar absorptivity

$b$  light path length

$c$  concentration of the fluorophore

$$I_f = N_f P_{abs} P_{em}$$

$N_f$  number of fluorophores in the light path

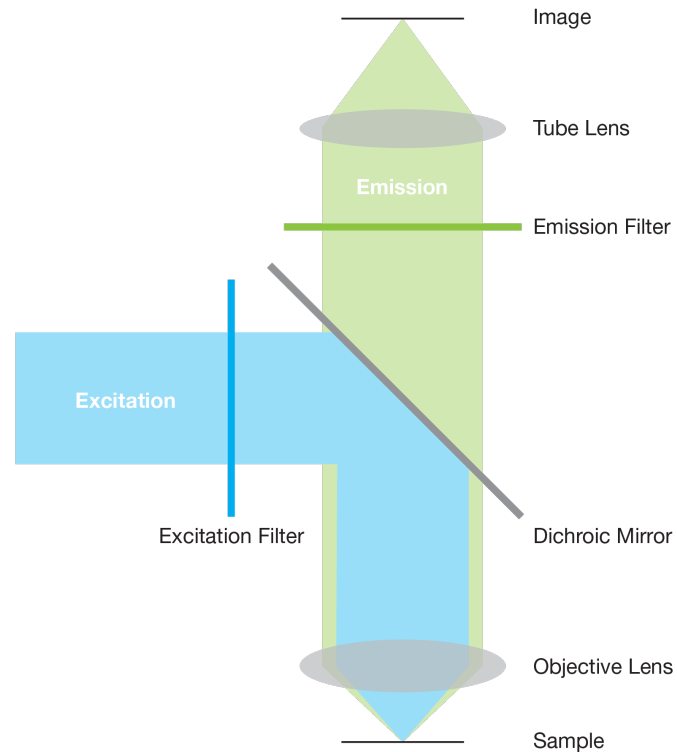
$$P_{abs} = f(\lambda_{ex}, I_o, k, b_{ex})$$

$$P_{em} = f(\lambda_{em}, k, b_{em})$$

# Fluorescence microscopy

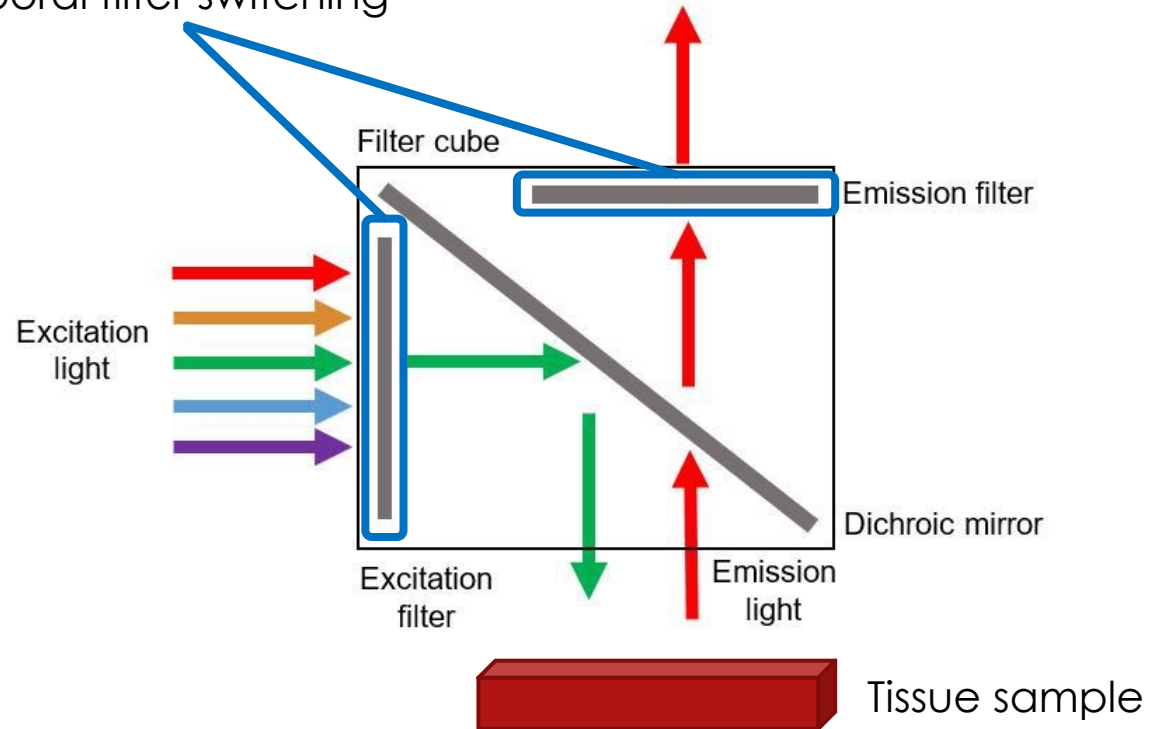
## THE LIGHT PATH

### Single-colour imaging



### Multi-colour(fluorophore) imaging

Temporal filter switching



# Fluorescence microscopy

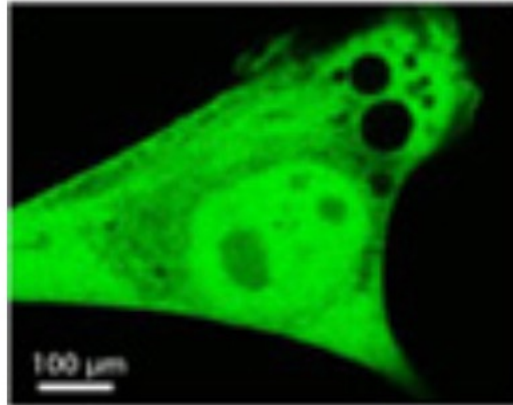
WHAT IS PHOTOBLEACHING?

Fluorescence microscopy

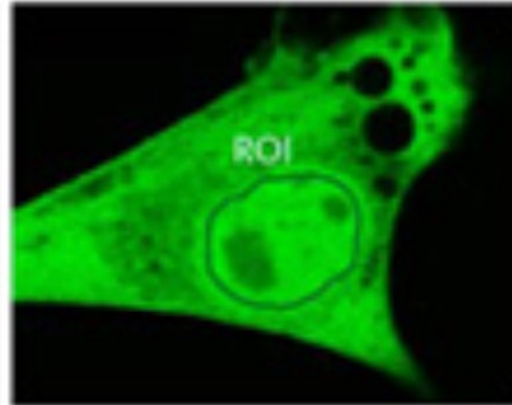
WHAT IS PHOTOBLEACHING?

# Fluorescence microscopy

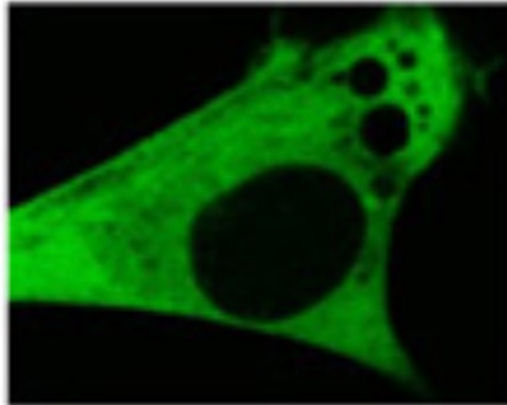
## WHAT IS PHOTOBLEACHING?



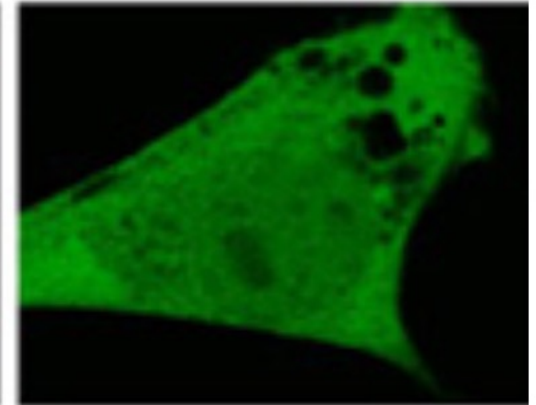
**Pre-bleach**



**Bleach ROI**



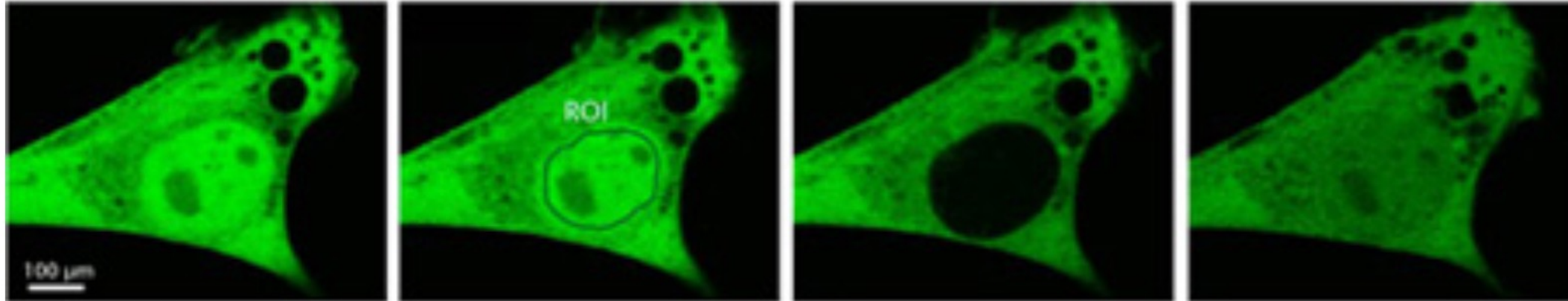
**Post-bleach**



**Recovered**

# Fluorescence microscopy

## WHAT IS PHOTBLEACHING?



Pre-bleach

Bleach ROI

Post-bleach

Recovered

Proportional to illumination time and  $I_0$

$$I_f = N_f P_{abs} P_{em}$$

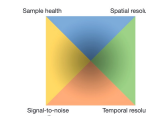


Figure 2 | The four main considerations for live imaging. This is also known as the 'pyramid of frustration', as no single parameter can be optimized without compromising the others.



# Fluorescence microscopy

## WHAT IS PHOTOTOXICITY?

[Published: 29 June 2017](#)

### **Assessing phototoxicity in live fluorescence imaging**

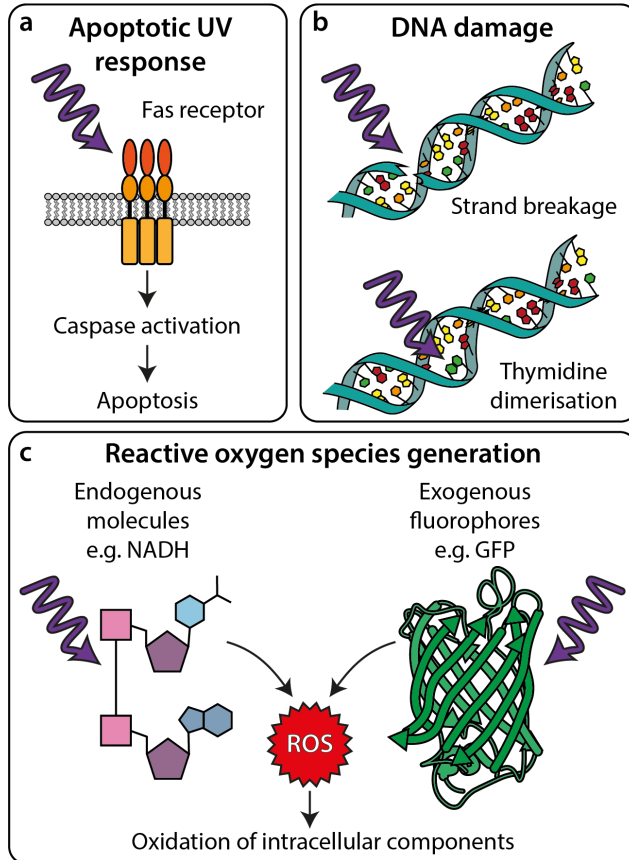
[P Philippe Laissue](#) , [Rana A Alghamdi](#), [Pavel Tomancak](#), [Emmanuel G Reynaud](#) & [Hari Shroff](#)

[Nature Methods](#) **14**, 657–661 (2017) | [Cite this article](#)

**8582** Accesses | **183** Citations | **47** Altmetric | [Metrics](#)

# Fluorescence microscopy

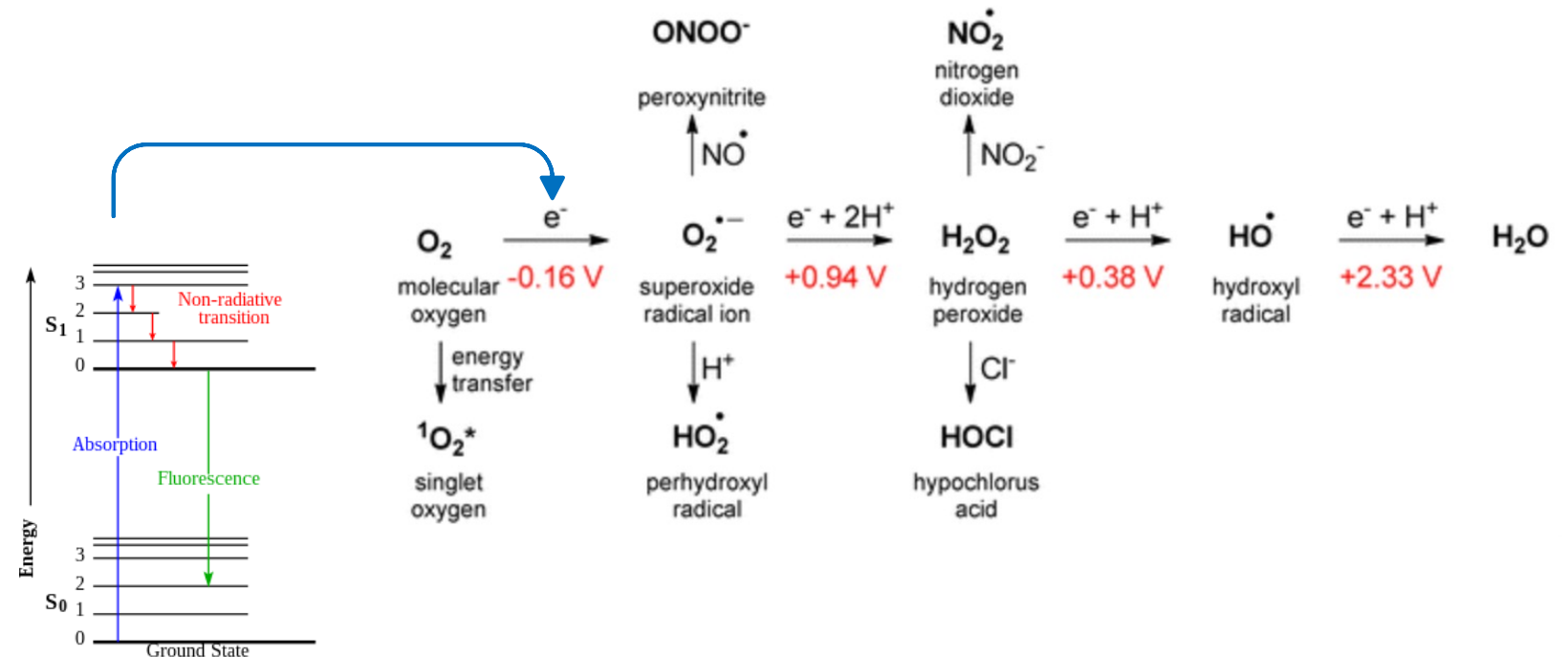
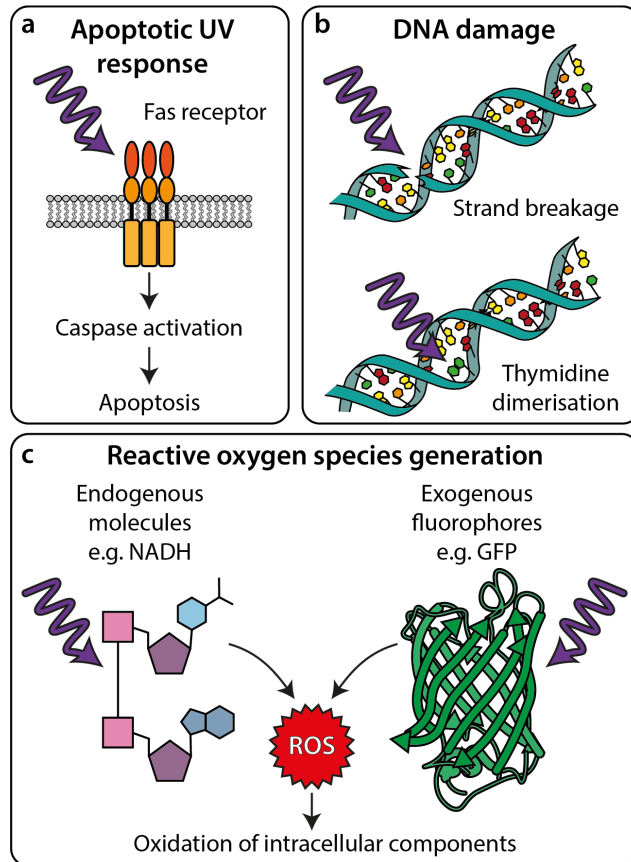
## WHAT IS PHOTOTOXICITY?



Zhou *et al.*, *Biochemistry*, 2021

# Fluorescence microscopy

## WHAT IS PHOTOTOXICITY?



Zhou *et al.*, *Biochemistry*, 2021

# Synthetic calcium fluorophores

## CALCIUM-BINDING IS STEP 1: WHAT ARE DESIRABLE CHARACTERISTICS?

Resting intracellular  $[\text{Ca}^{2+}]$  is  $\sim 100$  nM

Question: You have 1 teaspoon (5 g) of  $\text{CaCl}_2$ , how much water do you need to dissolve this in to achieve resting  $[\text{Ca}^{2+}]$ ?

# Synthetic calcium fluorophores

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Resting intracellular  $[\text{Ca}^{2+}]$  is  $\sim 100$  nM

Question: You have 1 teaspoon (5 g) of  $\text{CaCl}_2$ , how much water do you need to dissolve this in to achieve resting  $[\text{Ca}^{2+}]$ ?

Answer:  $\sim 450$  tonnes



# Synthetic calcium fluorophores

## CALCIUM-BINDING IS STEP 1: WHAT ARE DESIRABLE CHARACTERISTICS?

Question: What fraction of that resting 100 nM  $\text{Ca}^{2+}$  is free compared to bound to intracellular molecules?

# Synthetic calcium fluorophores

## CALCIUM-BINDING IS STEP 1: WHAT ARE DESIRABLE CHARACTERISTICS?

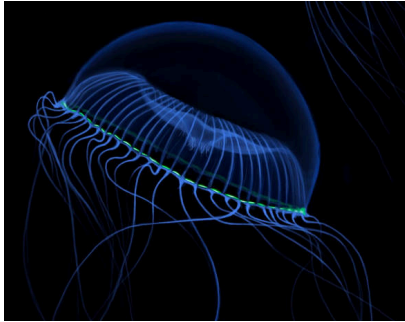
Question: What fraction of that resting 100 nM  $\text{Ca}^{2+}$  is free compared to bound to intracellular molecules?

Answer:  $\sim 1\%$

What is the implication for creating  $\text{Ca}^{2+}$ -sensitive fluorophores?

# Calcium-sensitive fluorophores

## ENDOGENOUS CALCIUM FLUOROPHORES



### Aequorea Victoria

Extraction, Purification and Properties of Aequorin,  
a Bioluminescent Protein from the Luminous  
Hydromedusan, *Aequorea*<sup>1</sup>

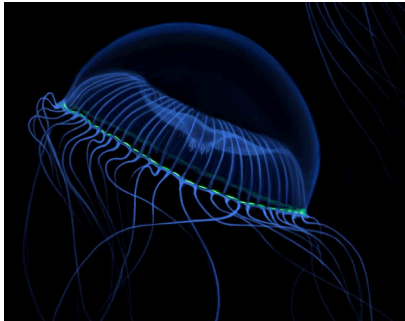
OSAMU SHIMOMURA,<sup>\*</sup> FRANK H. JOHNSON AND YO SAIGA  
Department of Biology, Princeton University, Princeton, New Jersey,  
and the Friday Harbor Laboratories, University of Washington,  
Friday Harbor, Washington

Shimomura *et al.* 1962

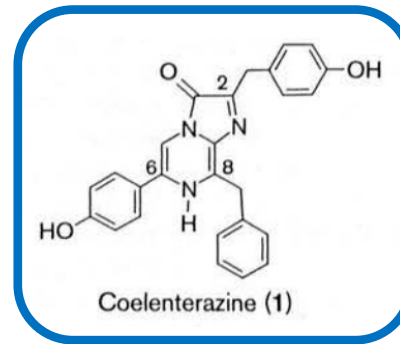
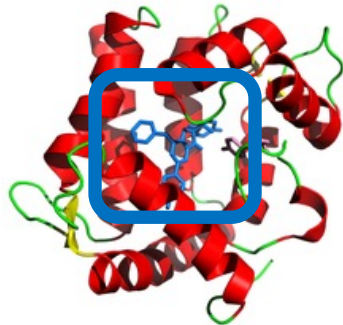
# Calcium-sensitive fluorophores

## ENDOGENOUS CALCIUM FLUOROPHORES

### Aequorin



*Aequorea Victoria*



Blue emitted light

Extraction, Purification and Properties of Aequorin,  
a Bioluminescent Protein from the Luminous  
Hydromedusan, *Aequorea*<sup>1</sup>

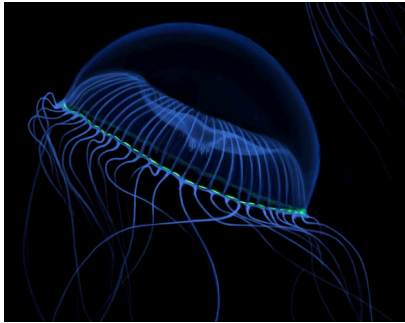
OSAMU SHIMOMURA,<sup>\*</sup> FRANK H. JOHNSON AND YO SAIGA  
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Shimomura *et al.* 1962

# Calcium-sensitive fluorophores

## ENDOGENOUS CALCIUM FLUOROPHORES

### Aequorin

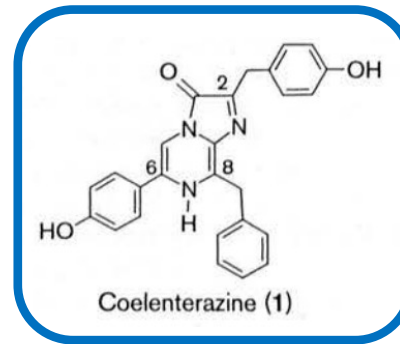
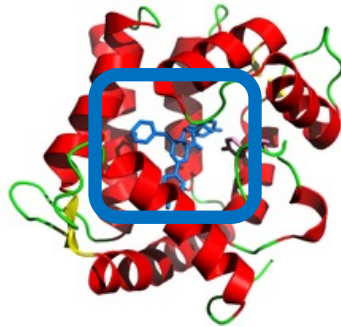


*Aequorea Victoria*

Extraction, Purification and Properties of Aequorin,  
a Bioluminescent Protein from the Luminous  
Hydromedusan, *Aequorea*<sup>1</sup>

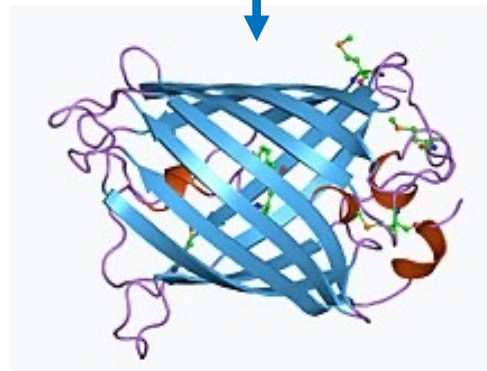
OSAMU SHIMOMURA,<sup>\*</sup> FRANK H. JOHNSON AND YO SAIGA  
Department of Biology, Princeton University, Princeton, New Jersey,  
and the Friday Harbor Laboratories, University of Washington,  
Friday Harbor, Washington

Shimomura *et al.* 1962



Blue emitted light

GFP



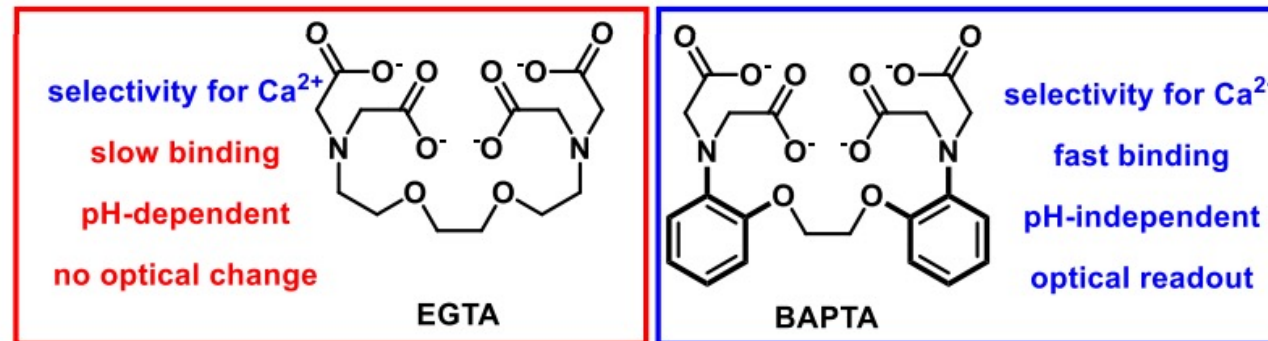
Green emitted light

Förster  
Resonance  
Energy  
Transfer

# Synthetic calcium fluorophores

## CALCIUM-BINDING IS STEP 1

BAPTA is the foundation of most synthetic  $\text{Ca}^{2+}$  indicators



Zhou et al., *Biochemistry*, 2021

### Pros

1.  $K_d = 100$  nM (perfect for biology)
2. pH-independent
3. Fast

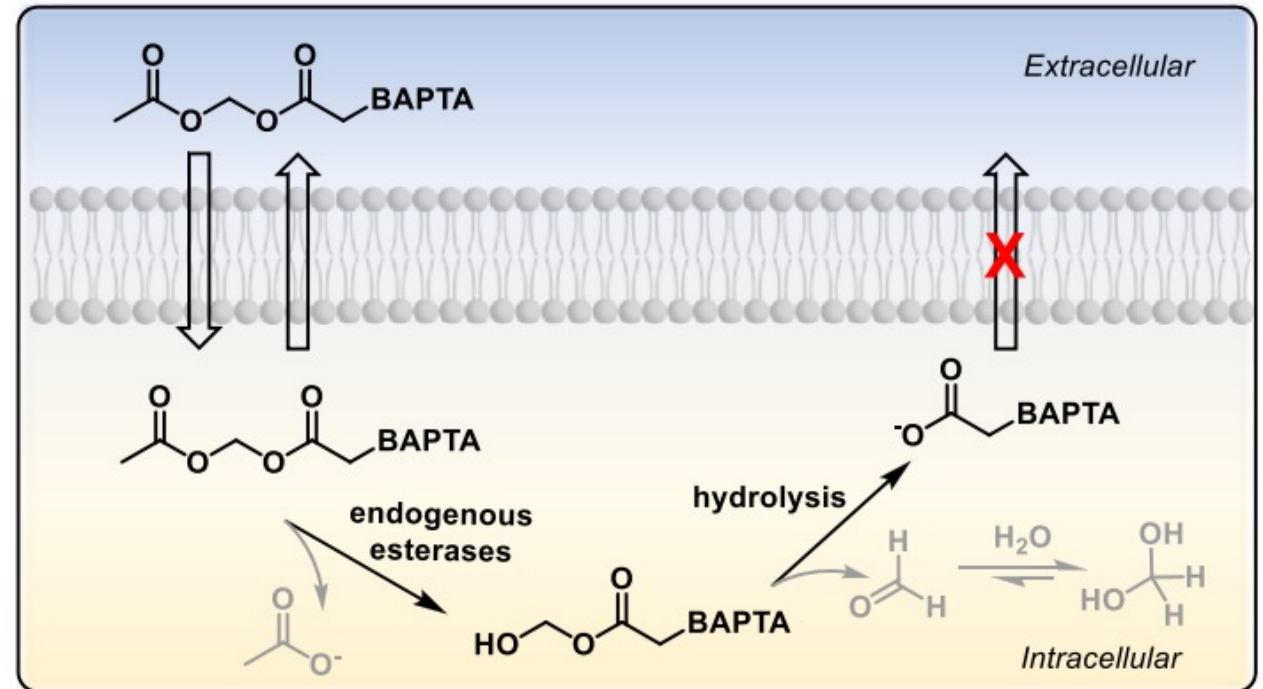
### Cons

1. Cell impermeant
2. Absorption peak  $\lambda \sim 200$  nm
3. Emission peak  $\lambda < 250$  nm

# Delivering calcium fluorophores

## HOW DO CALCIUM FLUOROPHORES GET INTO CELLS?

AM esters create cell-reversible lipid solubility



Zhou *et al.*, *Biochemistry*, 2021

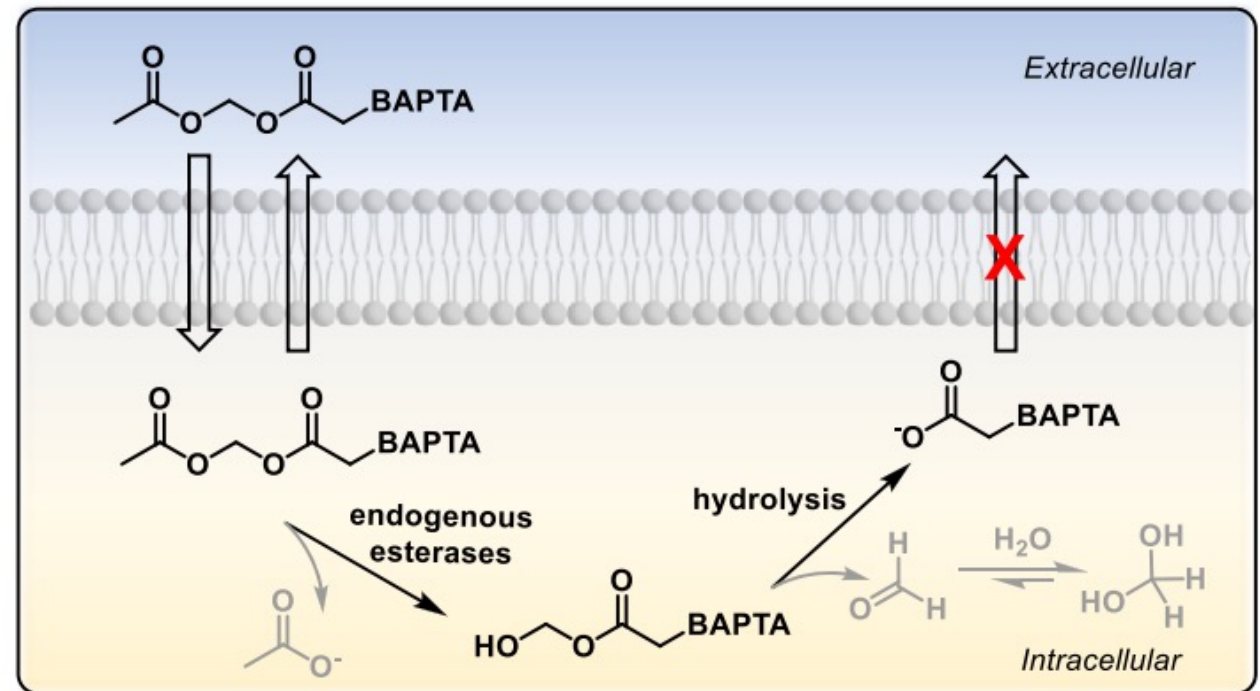


# Delivering calcium fluorophores

## HOW DO CALCIUM FLUOROPHORES GET INTO CELLS?

Issues with this mechanism?

AM esters create cell-reversible lipid solubility



Zhou *et al.*, *Biochemistry*, 2021

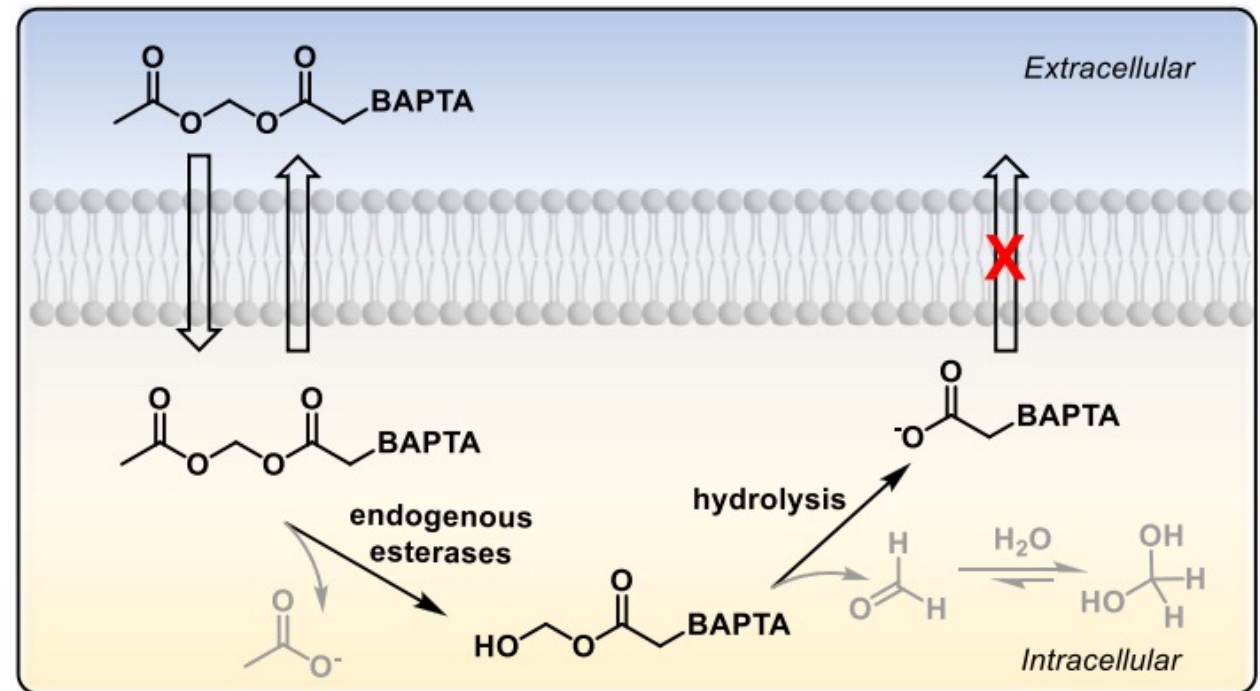
# Delivering calcium fluorophores

## CELL SOLUBLE CALCIUM FLUOROPHORES

AM esters create cell-reversible lipid solubility

### 1. Cell permeant

1. Adsorption to lipophilic substrates
2. Variable/regional fluorophore loading



Zhou *et al.*, *Biochemistry*, 2021

# Designing fluorescence

## RATIOMETRIC CALCIUM IMAGING

THE JOURNAL OF BIOLOGICAL CHEMISTRY  
© 1985 by The American Society of Biological Chemists, Inc.

Vol. 260, No. 6, Issue of March 25, pp. 3440-3450, 1985  
Printed in U.S.A.

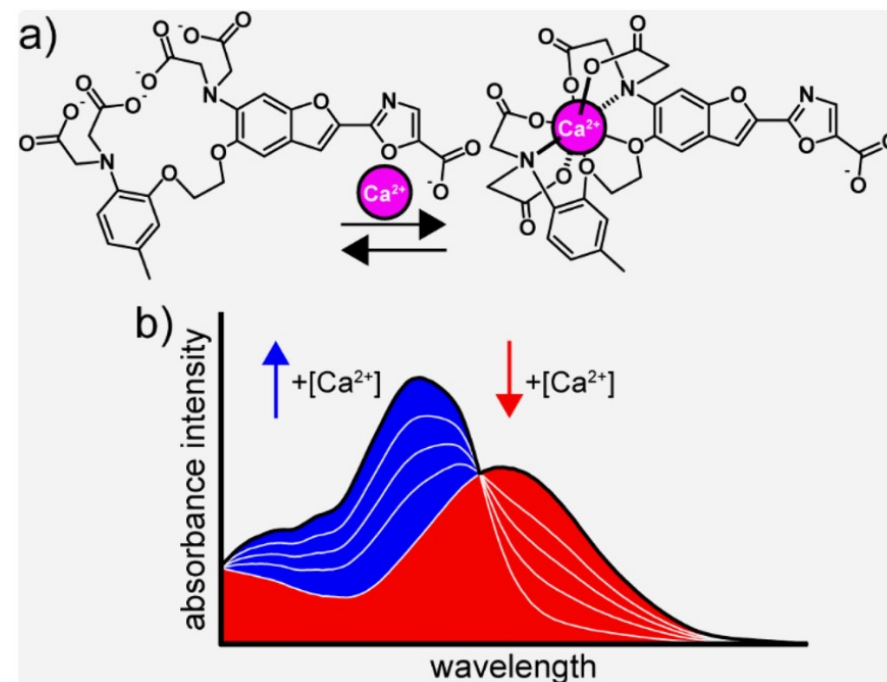
### A New Generation of $\text{Ca}^{2+}$ Indicators with Greatly Improved Fluorescence Properties\*

(Received for publication, August 23, 1984)

Grzegorz Grynkiewicz‡, Martin Poenie, and Roger Y. Tsien§

From the Department of Physiology-Anatomy, University of California, Berkeley, California 94720

### Benzofuran fluorophore



Zhou *et al.*, *Biochemistry*, 2021

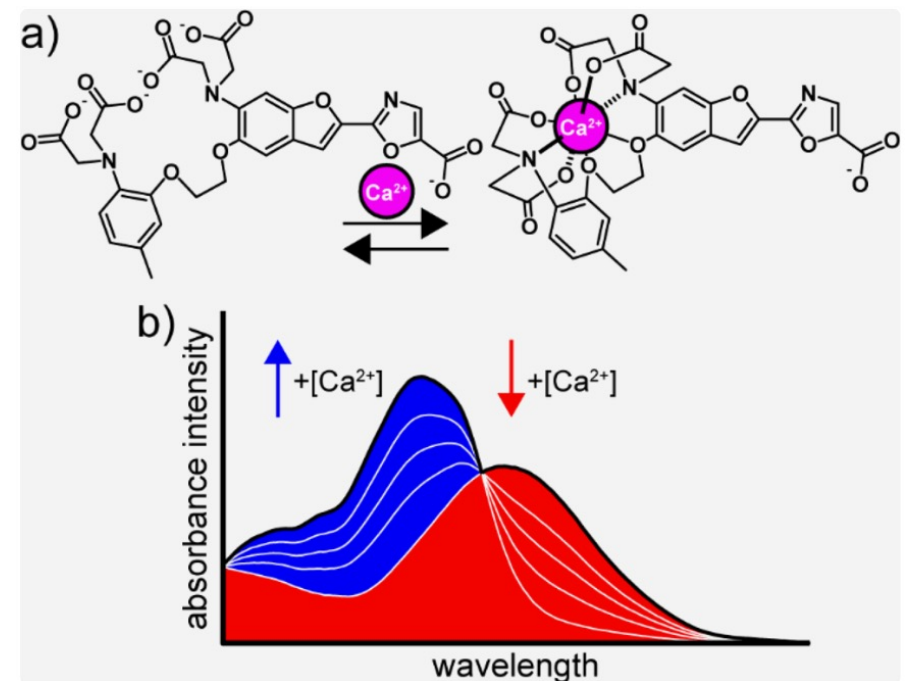
# Ratiometric calcium fluorophores

## RATIOMETRIC CALCIUM IMAGING

1. Cell permeant
2. Ratiometric  $\text{Ca}^{2+}$  imaging (Fura-2:  $F_{340}/F_{380}$ )
3. Visible emission
  - $\lambda_{\text{em}} = 510 \text{ nm}$

1. Modest quantum yield (0.5-0.6)
2. High energy UV excitation is toxic
3. Variable/regional fluorophore loading

### Benzofuran fluorophore



Zhou *et al.*, *Biochemistry*, 2021

# Ratiometric calcium fluorophores

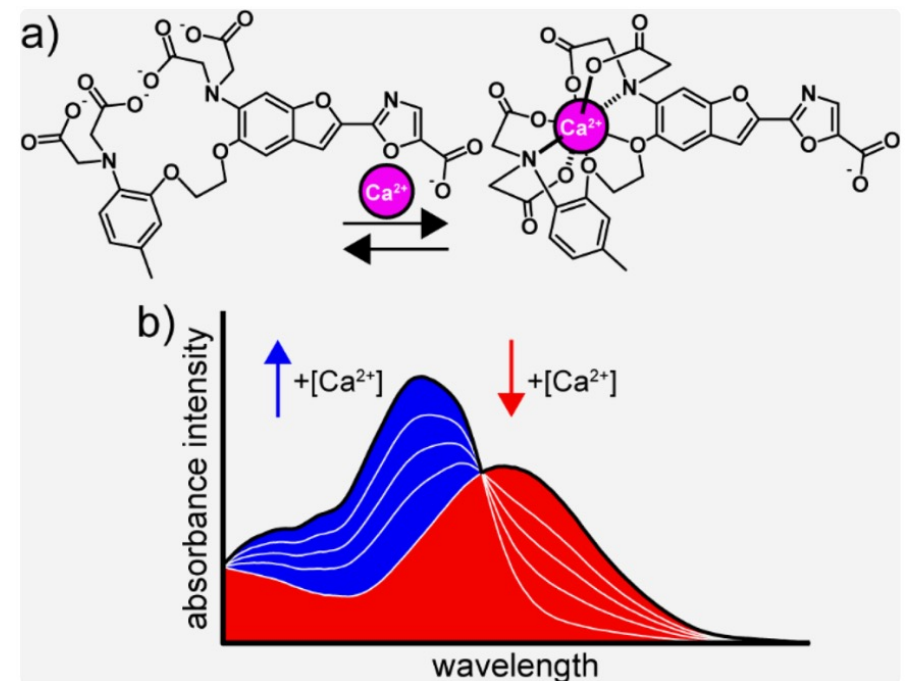
## RATIOMETRIC CALCIUM IMAGING

1. Cell permeant
2. Ratiometric  $\text{Ca}^{2+}$  imaging (Fura-2:  $F_{340}/F_{380}$ )
3. Visible emission
  - $\lambda_{\text{em}} = 510 \text{ nm}$

1. Modest quantum yield (0.5-0.6)
2. High energy UV excitation is toxic
3. Variable/regional fluorophore loading

Question: Advantage of ratiometric imaging?

### Benzofuran fluorophore



Zhou *et al.*, *Biochemistry*, 2021

# Ratiometric calcium fluorophores

## RATIOMETRIC CALCIUM IMAGING

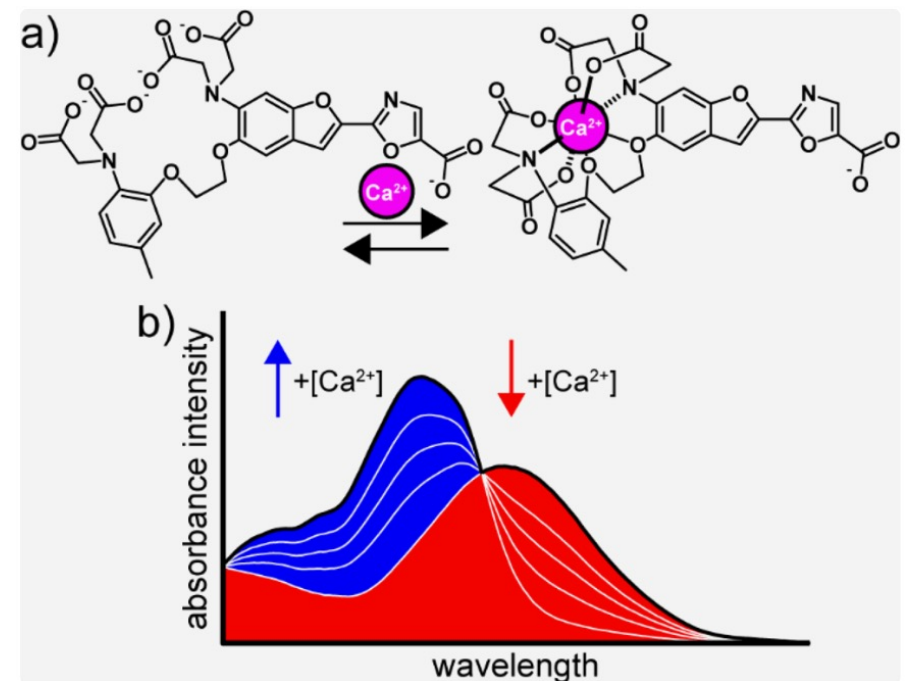
1. Cell permeant
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1. Modest quantum yield (0.5-0.6)
2. High energy UV excitation is toxic
3. Variable/regional fluorophore loading

Question: Advantage of ratiometric imaging?

Answer: Calibration

### Benzofuran fluorophore



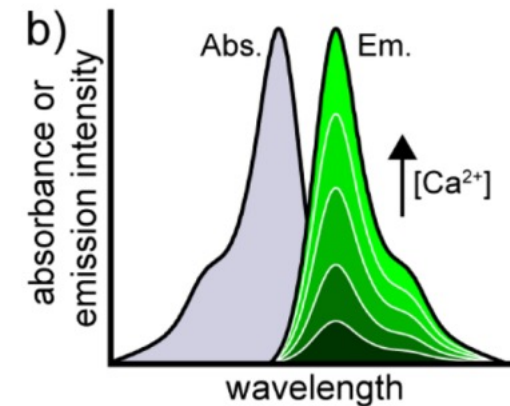
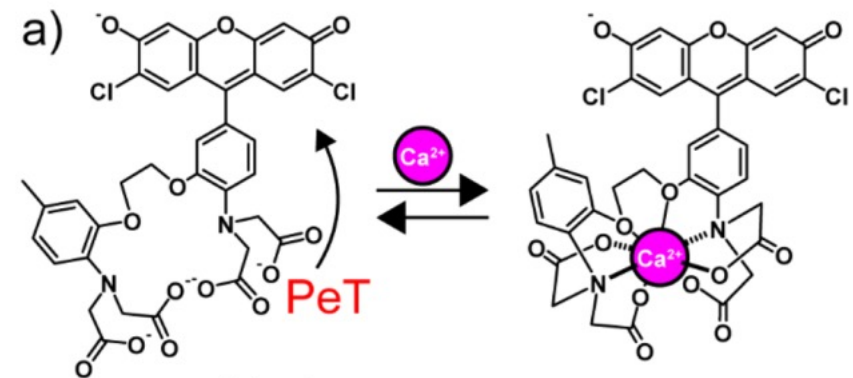
Zhou *et al.*, *Biochemistry*, 2021



# Intensity-based calcium imaging

## CALCIUM-DEPENDENT BRIGHTNESS

Photoinduced electron transfer (PeT) dyes



Zhou *et al.*, *Biochemistry*, 2021

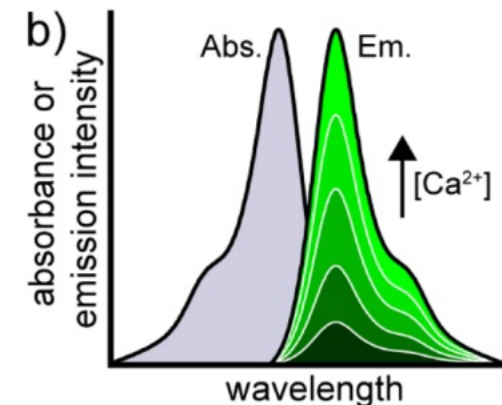
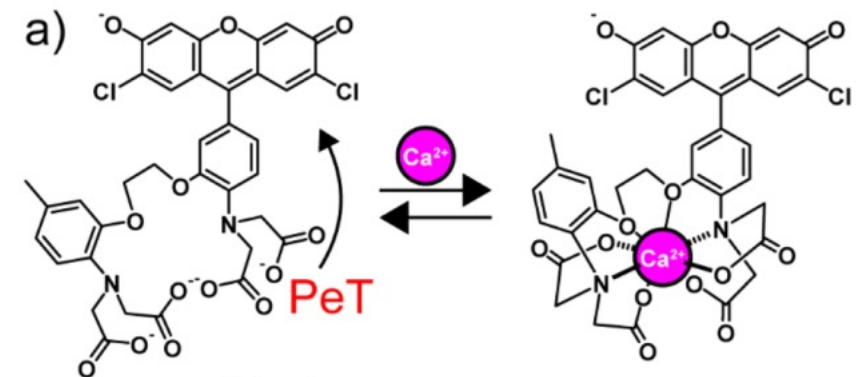


# Intensity-based calcium imaging

## CALCIUM-DEPENDENT BRIGHTNESS

1. Visible-range excitation
  2. High quantum yield – bright
  3. Multicolor emission
- 
1. Direct calibration not possible
  2. Variable/regional fluorophore loading

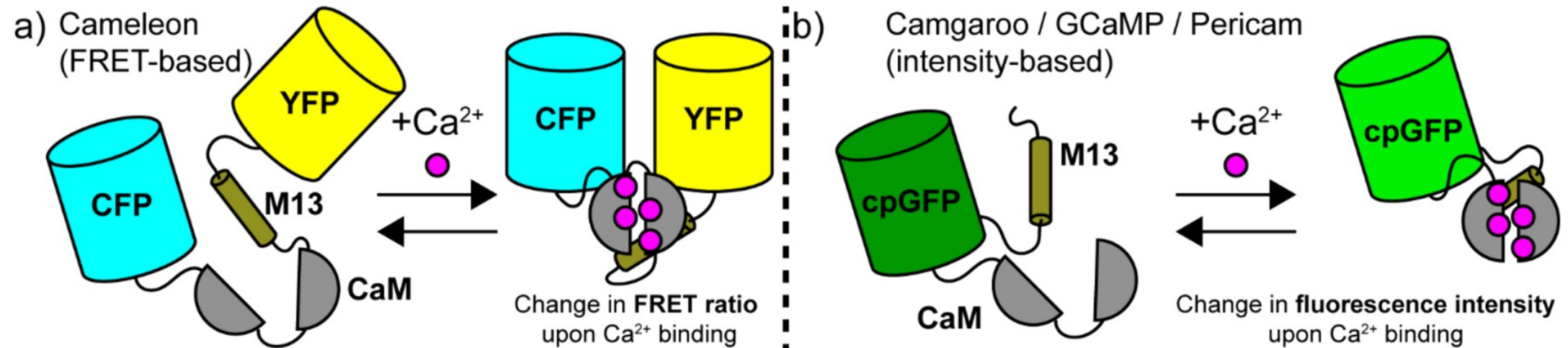
Photoinduced electron transfer (PeT) dyes



Zhou *et al.*, *Biochemistry*, 2021

# Leveraging protein fluorescence

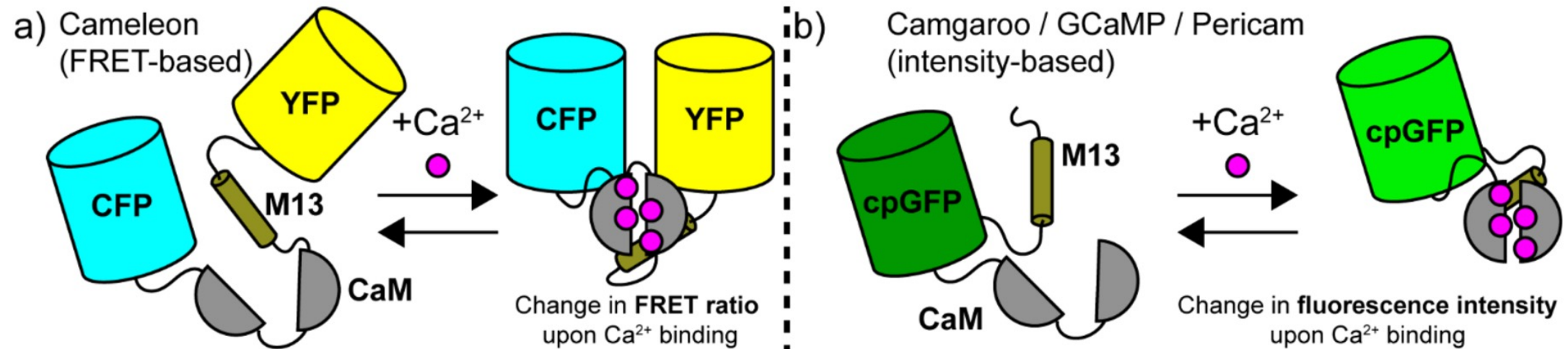
## GENETICALLY ENCODED CALCIUM INDICATORS (GECIS)



Zhou *et al.*, *Biochemistry*, 2021

# Leveraging protein fluorescence

## GENETICALLY ENCODED CALCIUM INDICATORS (GECIS)



Zhou et al., *Biochemistry*, 2021

### Pros

1. Much less variable fluorophore content
2. No spatial concerns with loading in tissue
3. Acute viral loading possible, but challenging

### Cons

1. Direct calibration still not possible
2. Dedicated cell lines required for reproducible *in vitro* designs

# Summary: Calcium fluorescence

## STATE OF THE ART FOR MPS MEASUREMENTS

### 1. Modern $\text{Ca}^{2+}$ dyes

- Bright
- Excellent dynamic range
- Quick loading (synthetic), or dedicated GECI cell lines
- Many colors (multiplexing)

### 2. Ongoing challenges

- Variable loading (synthetic dyes)
- Calibrated  $[\text{Ca}^{2+}]$  largely infeasible
- GECIs limited to specific engineered cell lines

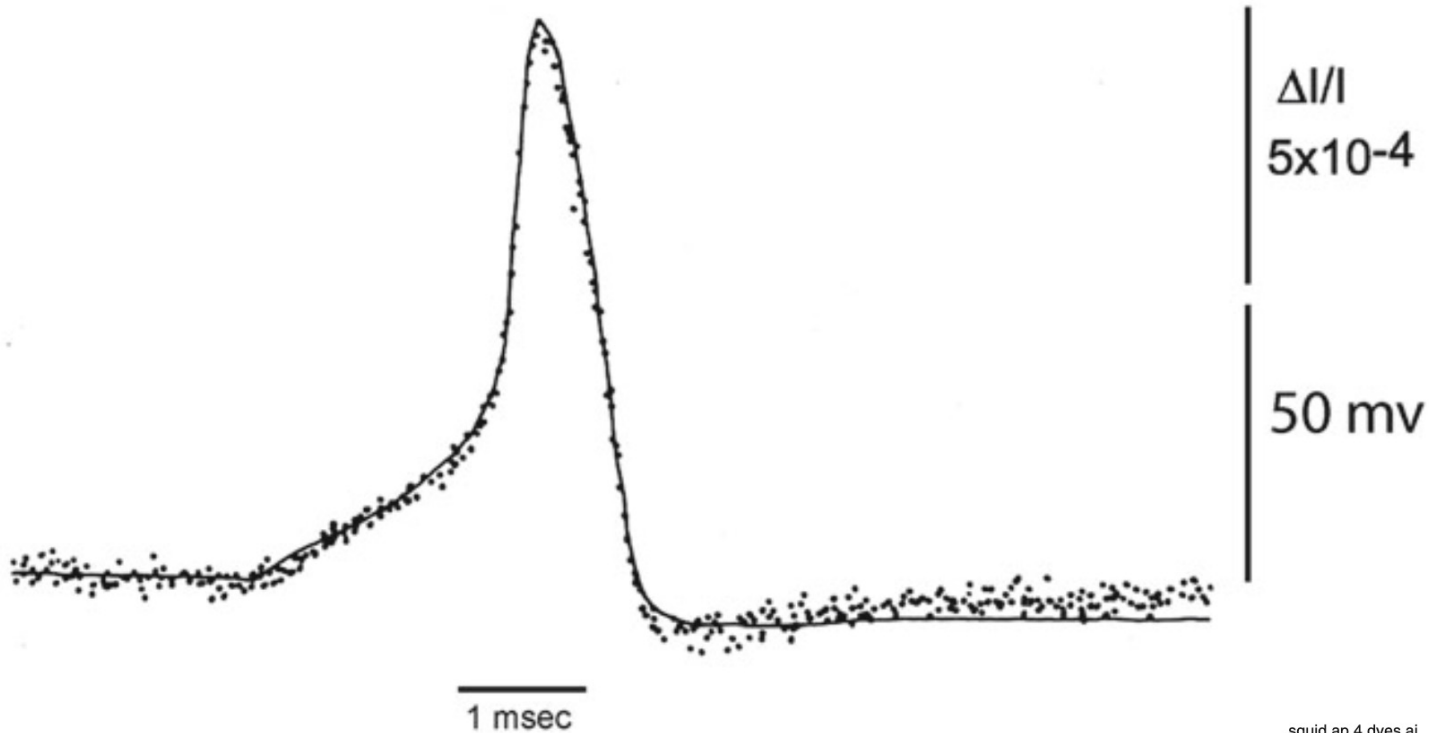
# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS

— 10 kHz electrode recording

∴ 1 kHz optical recording (50 sweeps)

SQUID AXON



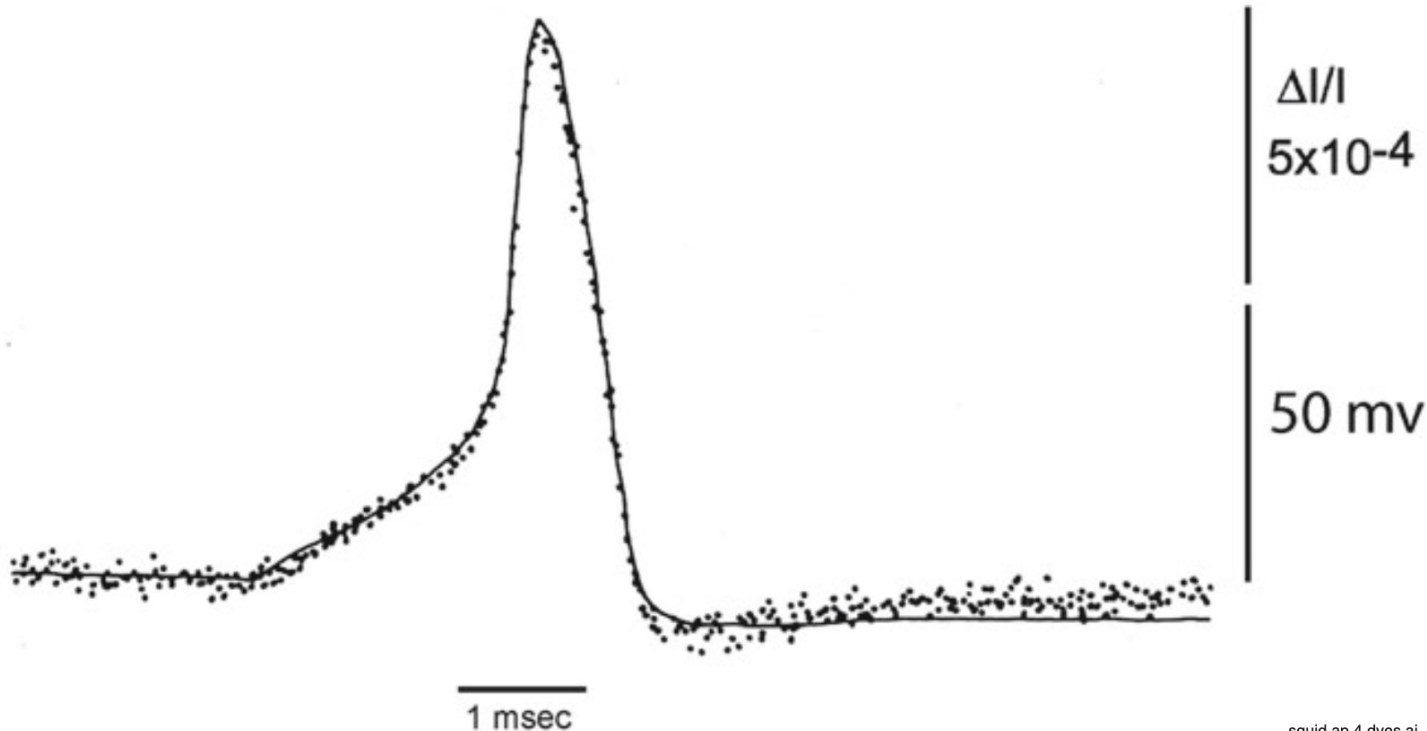
# Optical voltage sensors

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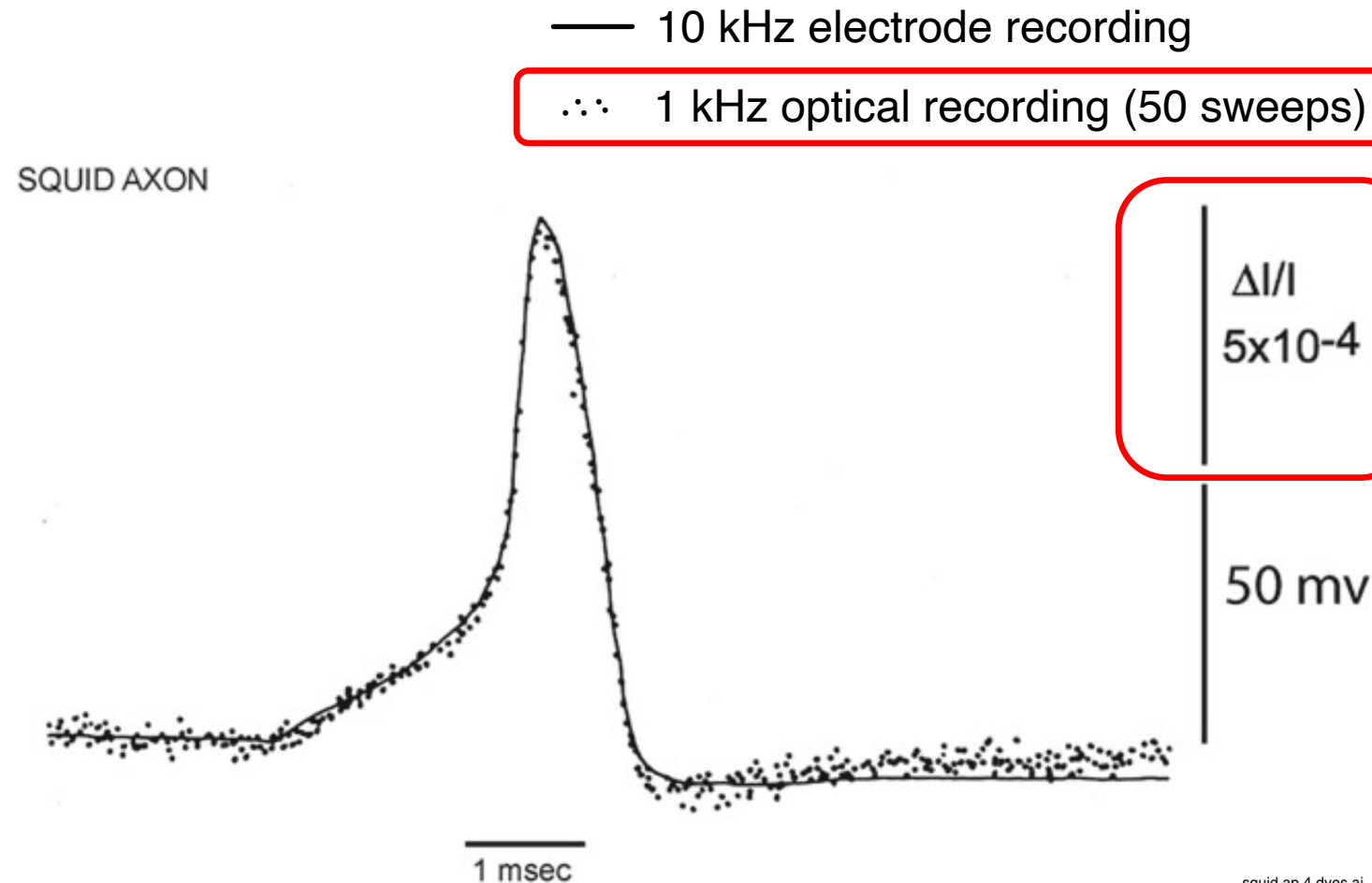
SQUID AXON



Question: Any fundamental differences compared to a calcium recording?

# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS



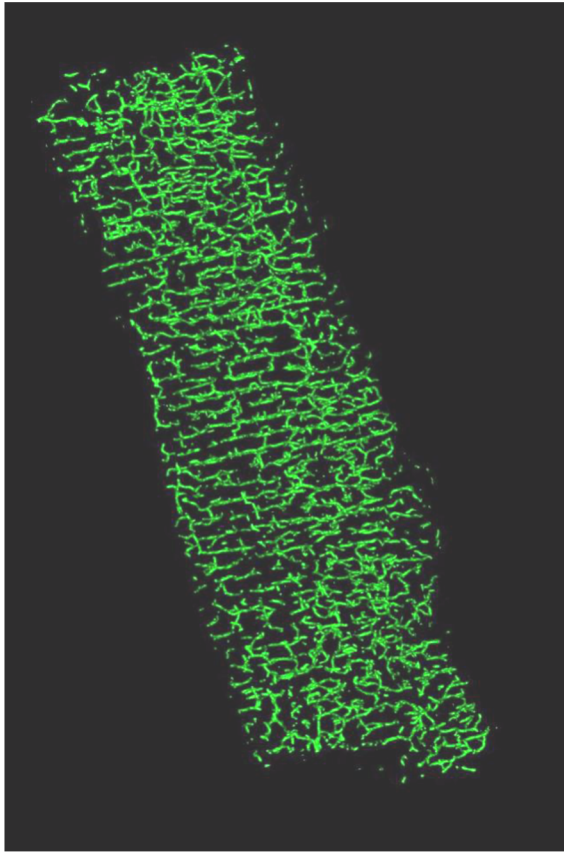
Question: Any fundamental differences compared to a calcium recording?

Answer: Fast voltage responses and low fluorophore density challenge signal-to-noise



# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS



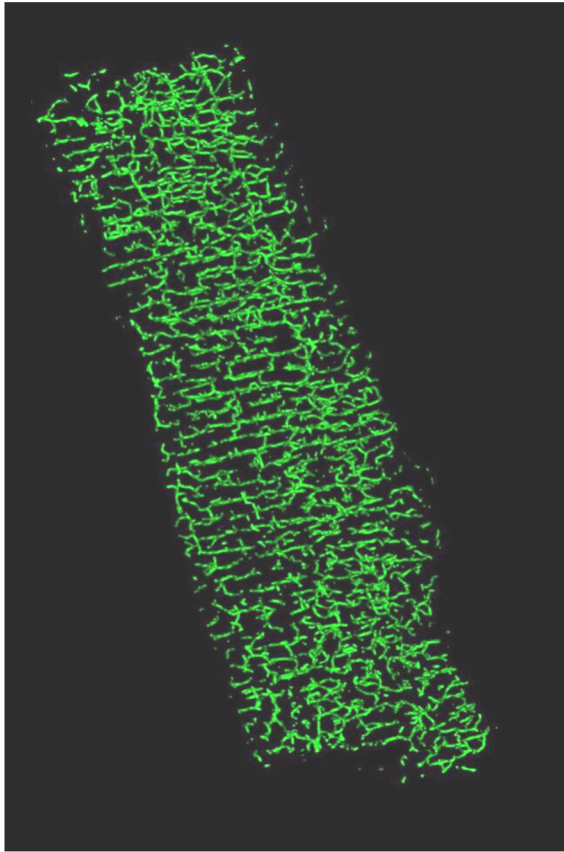
Cardiac cell cytosol volume:

$$W \times H \times L = 10 \times 20 \times 100 \text{ } (\mu m^3) = 20 \text{ pL}$$

Question: Typical intracellular concentration for calcium fluorophores?

# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS



Cardiac cell cytosol volume:

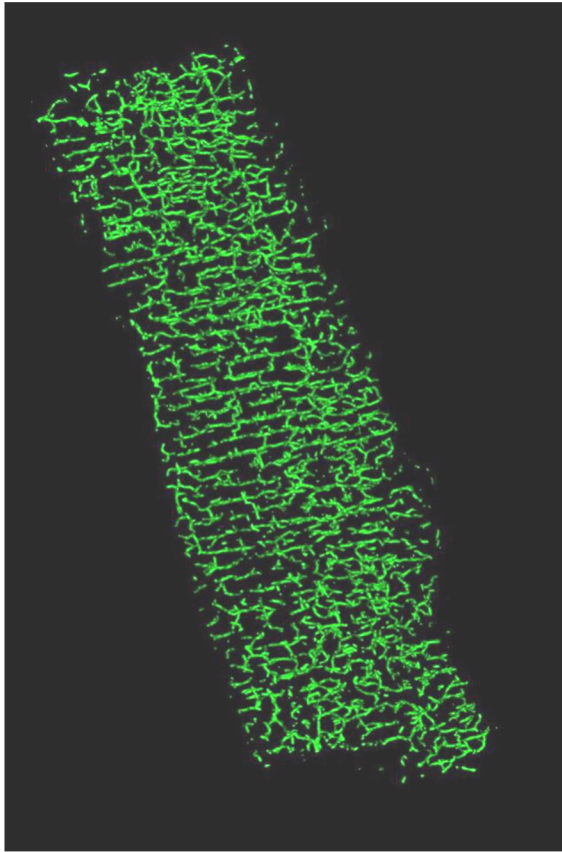
$$W \times H \times L = 10 \times 20 \times 100 (\mu m^3) = 20 pL$$

Question: Typical intracellular concentration for calcium fluorophores?

Answer:  $\sim 200 \mu M$

# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS



Cardiac cell cytosol volume:

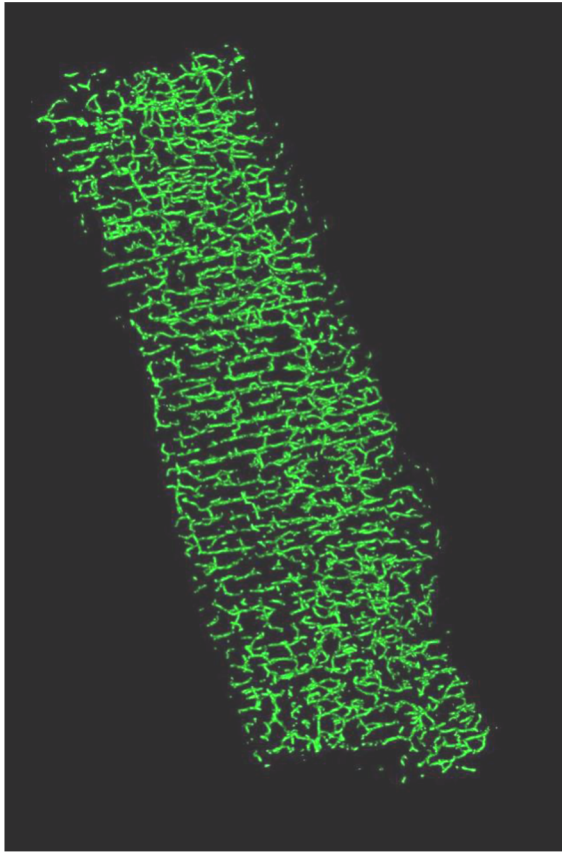
$$W \times H \times L = 10 \times 20 \times 100 (\mu m^3) = 20 pL$$

Cardiac cell membrane volume:

$$\frac{C (\mu F)}{C_m \left( \frac{\mu F}{cm^2} \right)} \times W_m (cm) = \frac{2 \times 10^{-4}}{1} \times 2 \times 10^{-7} (cm^3) = 40 fL$$

# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS



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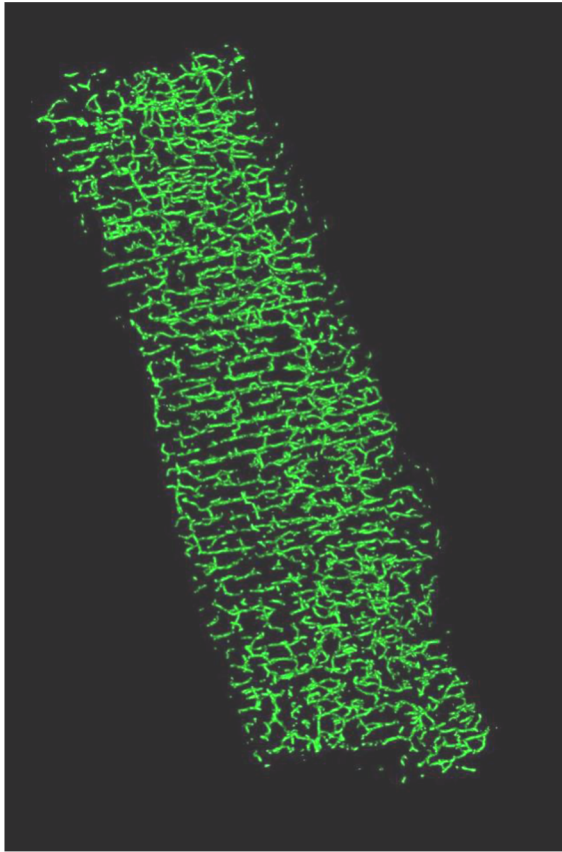
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Question: Implication for simultaneous  $V_m$ - $Ca^{2+}$  imaging ( $I_f = N_f P_{abs} P_{em}$ )?

# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS



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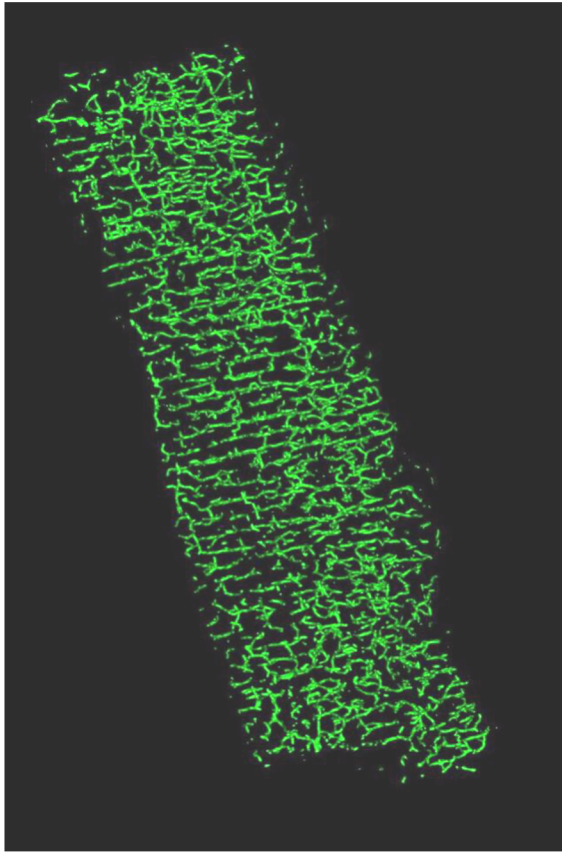
Question: Implication for simultaneous  $V_m$ - $Ca^{2+}$  imaging ( $I_f = N_f P_{abs} P_{em}$ )?

Answer:  $I_0$  required for high fidelity  $V_m$  imaging is much more than for  $Ca^{2+}$

= major constraints on experimental design

# Optical voltage sensors

## ADDED CHALLENGES FOR OPTICAL VOLTAGE SENSORS



Cardiac cell cytosol volume:

$$W \times H \times L = 10 \times 20 \times 100 (\mu m^3) = 20 pL$$

Cardiac cell membrane volume:

$$\frac{C (\mu F)}{C_m \left( \frac{\mu F}{cm^2} \right)} \times W_m (cm) = \frac{2 \times 10^{-4}}{1} \times 2 \times 10^{-7} (cm^3) = 40 fL$$

Question: Can we simply enrich the membrane with  $V_m$  fluorophores?

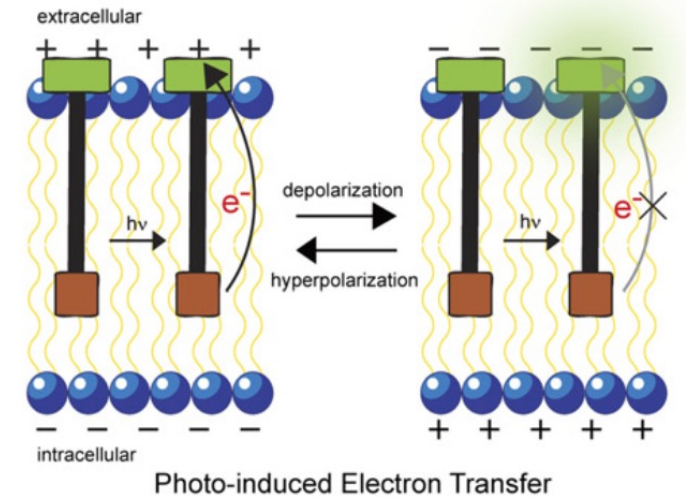
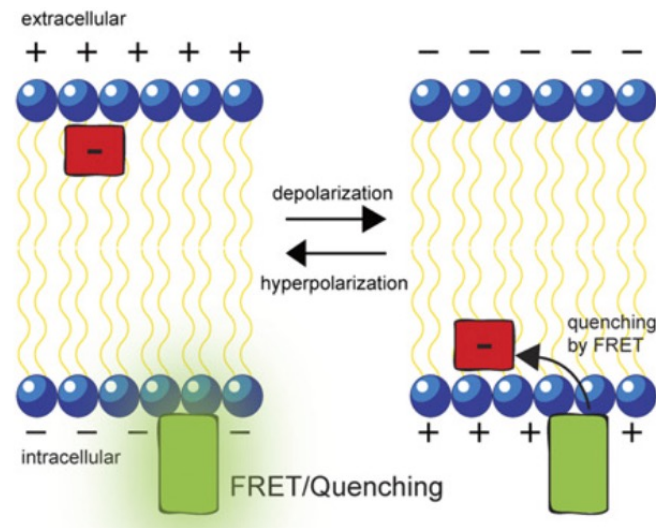
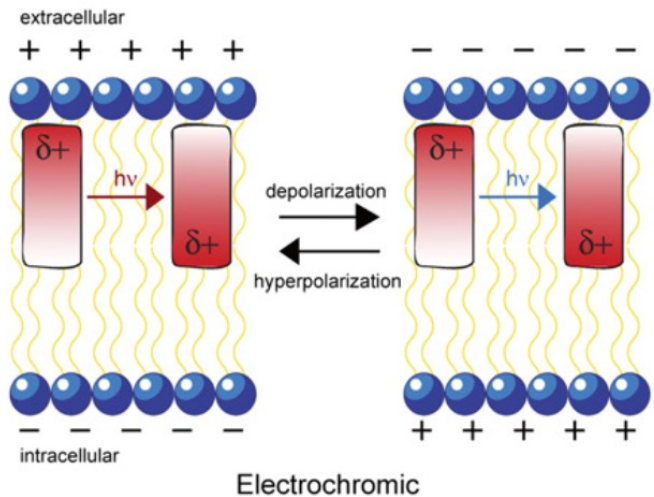


# Optical voltage sensors

## STRATEGIES FOR DEVELOPING OPTICAL VOLTAGE SENSING

Question: Can we simply enrich the membrane with  $V_m$  fluorophores?

Answer: Not really, all  $V_m$  fluorophores embed in membranes and act as detergents





100

The diagram illustrates three mechanisms of voltage-dependent fluorescence:

- Electrochromic:** Shows a lipid bilayer with a fluorophore (red rectangle) that changes its absorption spectrum ( $\delta+$ ) upon depolarization (hyperpolarization). The mechanism is driven by the electric field across the membrane. Below the diagram are three fluorescence traces at different membrane potentials: -90 mV (red shaded area), -45 mV (black trace), and 0 mV (black trace). The traces show a shift in the fluorescence peak and a change in intensity, labeled  $\Delta F/F$ .
- FRET/Quenching:** Shows a lipid bilayer with a fluorophore (red rectangle) and a quencher (green rectangle). The mechanism involves energy transfer (FRET) from the fluorophore to the quencher, leading to quenching. The diagram shows the transition between depolarized and hyperpolarized states, with the quencher being more effective in the hyperpolarized state.
- BeRST (Photo-induced Electron Transfer):** Shows a lipid bilayer with a fluorophore (green rectangle) and a quencher (red rectangle). The mechanism involves photo-induced electron transfer (PET) from the fluorophore to the quencher, leading to quenching. The diagram shows the transition between depolarized and hyperpolarized states, with the quencher being more effective in the hyperpolarized state.

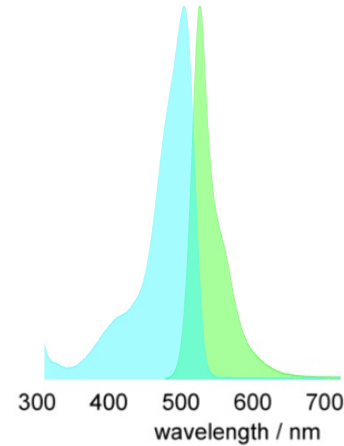
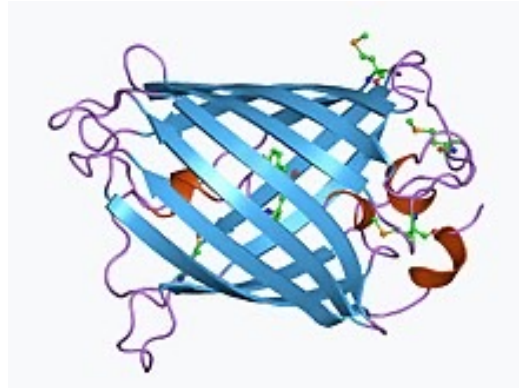
Miller *et al.* 2011

- High quantum yield
- Fast
- Far-red shifted
- Calibration not possible
- Not yet broadly distributed

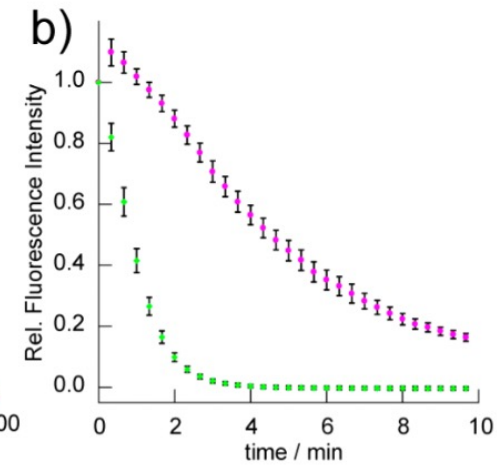
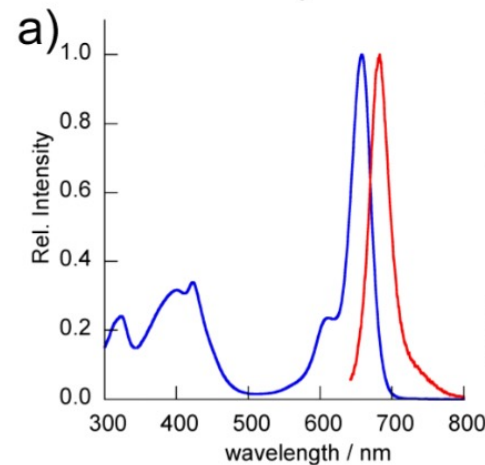
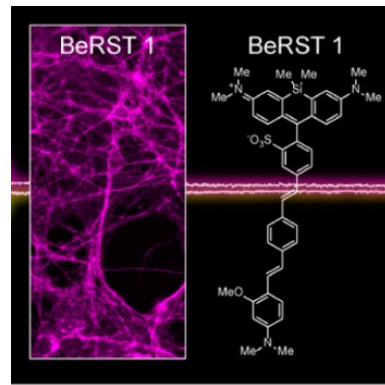
# Combining $V_m$ and $Ca^{2+}$ sensors

## SPECTRAL DISTINCTION AND PHOTOSTABILITY

GFP



BeRST



# Summary: Voltage fluorescence

## STATE OF THE ART FOR MPS MEASUREMENTS

### 1. Modern $V_m$ dyes

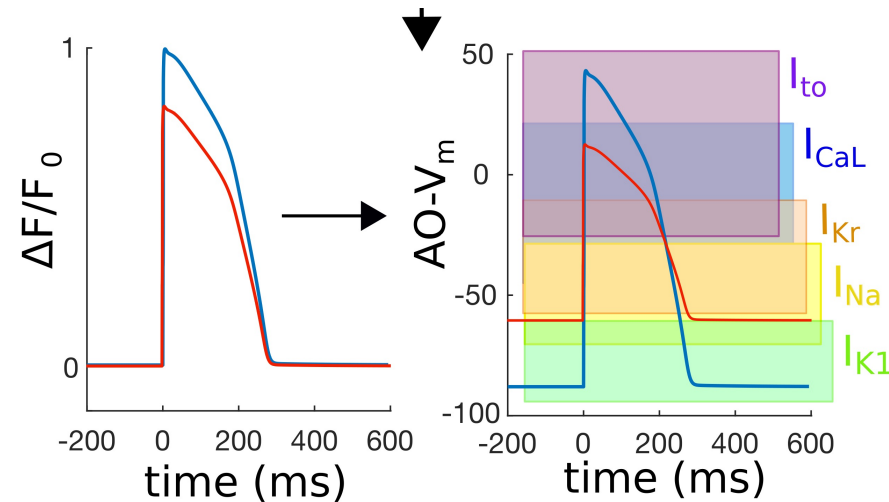
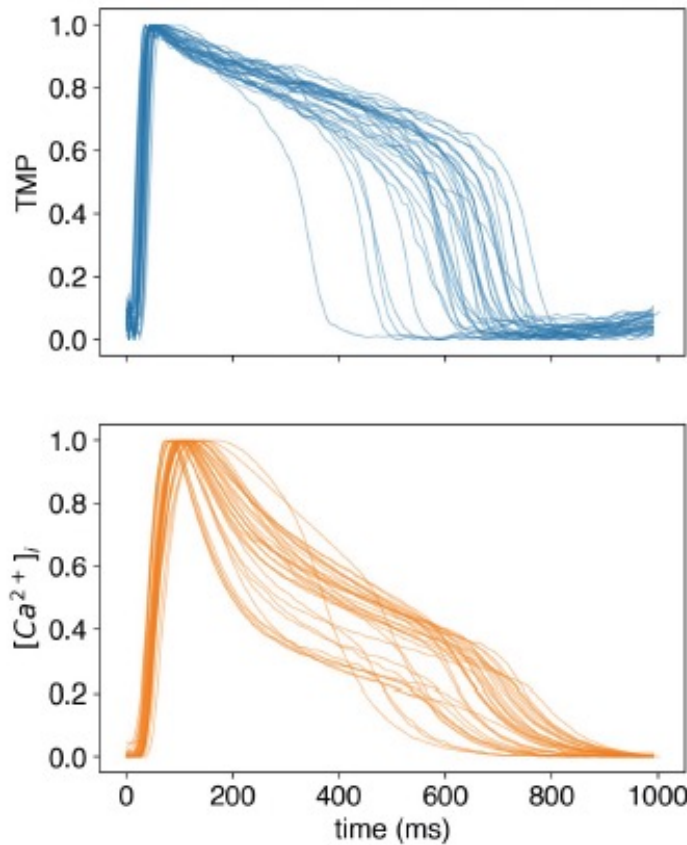
- Red-shifted spectra available
- Fast
- Improved  $\Phi$  through PeT and electrochromic GEVIs

### 2. Ongoing challenges

- Reconciling temporal constraints and signal strength
- High illumination intensity light required
- Calibration remains a challenge

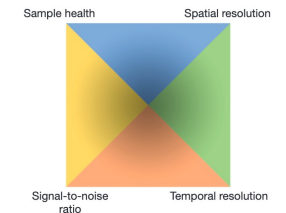
# Closing arguments: Calibration

## STATE OF THE ART FOR MPS MEASUREMENTS



Rich protocols:

- Multiple [drug]
- Frequency
- Extracellular [ion]



Calibrated signals